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**ASSEMBLY OF THE SOUTHERN MACROCYCLIC HALF OF  
(+)-SPIRASTRELOLIDE A THROUGH CYCLIC ACETAL TETHERED  
RING-CLOSING METATHESIS AND 1,3-ANTI-MUKAIYAMA-ALDOL†**

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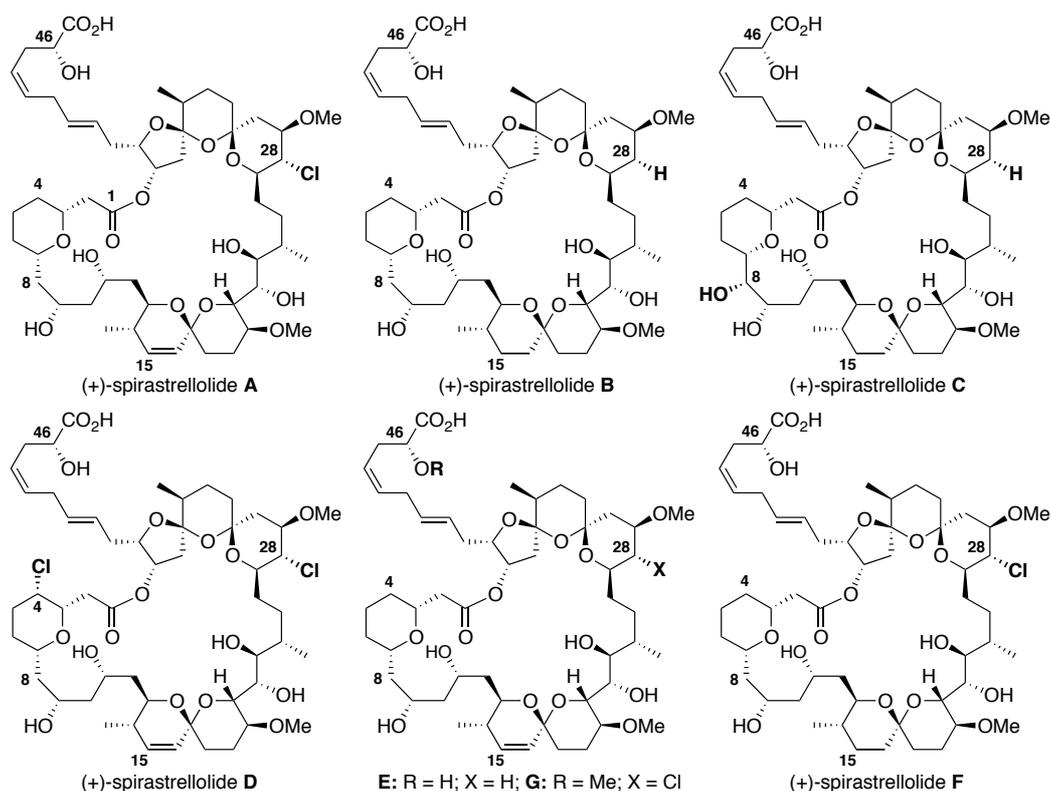
**Abstract** – We describe herein details of our efforts in syntheses of A-ring and BC-ring of (+)-spirastrellolide A. While the former would constitute a facile 12-step synthetic endeavor starting from 1,5-pentanediol, the latter would showcase a cyclic acetal-tethered ring-closing metathesis [RCM] method that was developed in our lab for *de novo* synthesis of spiroketals. Constructing the entire Southern Half of the macrocycle would require 1,3-*anti*-Mukaiyama aldol addition for connecting A-ring and BC-ring specifically at C10 and C11, thereby culminating a 17-step approach for the Southern Macrocyclic Half linearly from (+)-2,3-(*O*)-iso-propylidene-L-threitol. Also discussed here is the possibility of pursuing a more convergent approach toward the assembly of the Southern Half through first connecting A-ring and C-ring via acetal formation that would first link together the free C13-OH with C17 at the spiro-BC-ring junction. An ensuing application of our cyclic acetal-tethered RCM strategy to close B-ring would adopt this cyclic acetal intermediate.

†This paper is dedicated to Professor Ei-Ichi NEgishi with the deepest respect in honoring the special occasion of his 77<sup>th</sup> birthday.

## INTRODUCTION

Roberge and Andersen *et al.* in 2003 unveiled the rough structure of (+)-spirastrellolide A, a macrocyclic

lactone that is rich in spiroketal motifs.<sup>1a</sup> Subsequently in 2004, they revised and completed the structural assignment of (+)-spirastrellolide A including all relative stereochemistry sans C46.<sup>1b</sup> In 2007, with the isolation of (+)-spirastrellolide B, the absolute configuration of the macrocyclic core was established through X-ray crystallography.<sup>1c</sup> In the same year, (+)-spirastrellolide C-G were also identified from the same marine sponge *Spirastrella coccinea*.<sup>1d</sup> (+)-Spirastrellolide A possess the ability to initiate premature entry into mitosis and untimely mitotic arrest in cells, and more importantly, it exhibits a potent inhibitory activity against protein phosphatase 2A [ $IC_{50} = 1 \text{ nM}$ ]. The level of inhibitory selectivity is also outstanding in favor of PP2A over PP1 by a factor of 50 in terms of  $IC_{50}$ .<sup>2</sup> In addition, (+)-spirastrellolide A does not inhibit PP2C! Its biological activities, therefore, resemble other known Ser/Thr phosphatase inhibitors fostriecin and okadaic acid.<sup>3</sup>



**Figure 1.** *Spirastrellolides A-G.*

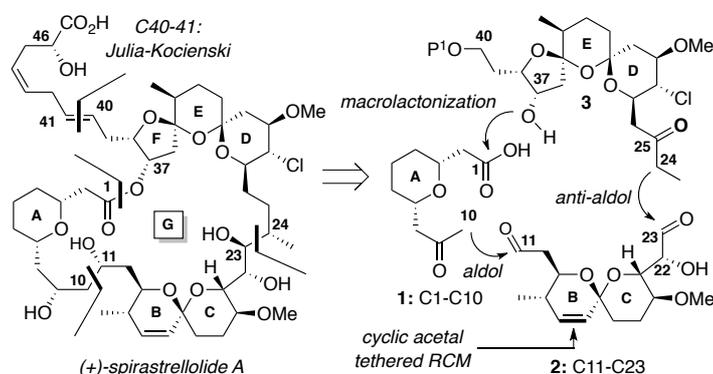
Development of protein phosphatase inhibitors has lagged behind the interest in kinase inhibitors because of the perceived notion that kinases are more highly regulated and specific.<sup>4</sup> However, there has been a renewed interest in recent years because reversible protein phosphorylation is critical "as the other half" of checkpoints in cell cycles, and protein phosphatases assume an equally important role in regulating cellular signal transductions and should not be ignored. Designing phosphatase inhibitors can lead to new paradigms in developing cancer therapeutics.<sup>4</sup> As a result, this family of natural products has attracted an

elegant array of synthetic efforts<sup>5</sup> with the very first total synthesis being reported by Paterson.<sup>6</sup> We became interested in spirastrellolide A because we have been developing cyclic acetal tethered methods<sup>7a-c</sup> such as RCM<sup>7d-g,8</sup> as an unconventional approach to constructing spiroketals.<sup>9</sup> We report here our efforts toward the Southern Half of macrocyclic core of (+)-spirastrellolide A.

## RESULTS AND DISCUSSION

### 1. Retrosynthetic Analysis.

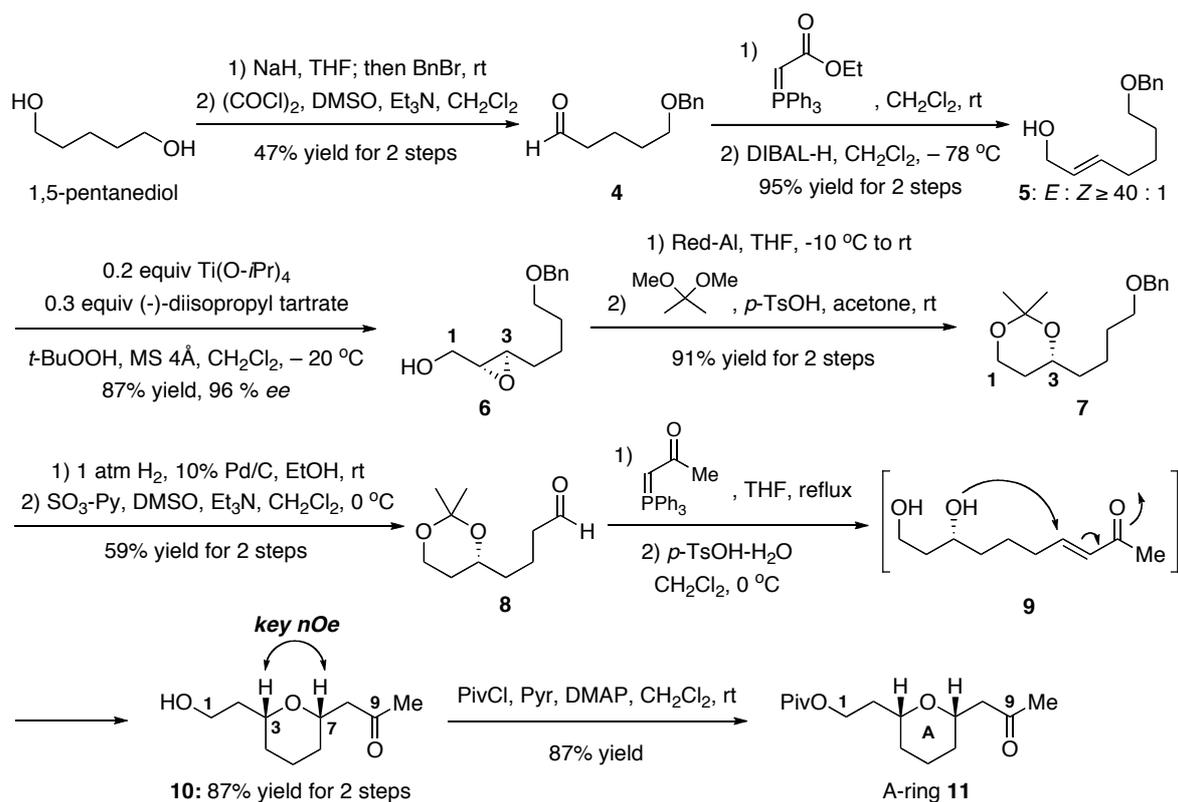
Our synthetic analysis would first involve a disconnection at C40-41 that will be reconnected through Julia-Kocienski olefination, which was also documented in Smith<sup>5j</sup> [**Scheme 1**]. Consequently, this would leave behind the central macrocyclic core [labeled as G-ring] that can be further divided into three major fragments: **1** [C1 through C10], **2** [C11 through C23], and **3** [C24 through C40]. A macrolactonization could link C1 and C37 between fragments **1** and **3**. A methyl ketone enolate *anti*-aldol will be used to connect fragments **1** and **2** at C10 and C11,<sup>5e,10</sup> and another *anti*-aldol would link together fragments **2** and **3** at C23 and C24 but will require the addition of the C25-oxo group. Synthesis of fragment **2** could feature our cyclic acetal tethered RCM.<sup>7d-g</sup>



**Scheme 1.** An Overview of Retrosynthetic Plan for Spirastrellolides A.

### 2. A-Ring Synthesis.

Our synthesis of A-ring<sup>10</sup> would call for the commercially readily available 1,5-pentanediol as the starting point. As shown in **Scheme 2**, 1,5-pentanediol could be quickly transformed into aldehyde **4** in 47% overall yield via mono-benzoylation and standard Swern oxidation. Standard HEW-modified Wittig-olefination followed by DIBAL-H reduction afforded exclusively *E*-allylic alcohol **5** in 95% overall yield. Sharpless asymmetric epoxidation employing D-(-)-diisopropyl tartrate provided epoxy alcohol **6** in 96% enantiomeric excess. It is noteworthy that we have provided here a completely different approach for establishing the C3 stereochemistry in A-ring of (+)-spirastrellolide A, as both Paterson<sup>5a-d,6</sup> and De Brabander<sup>5g</sup> employed an asymmetric Brown-allylation.<sup>11a</sup>



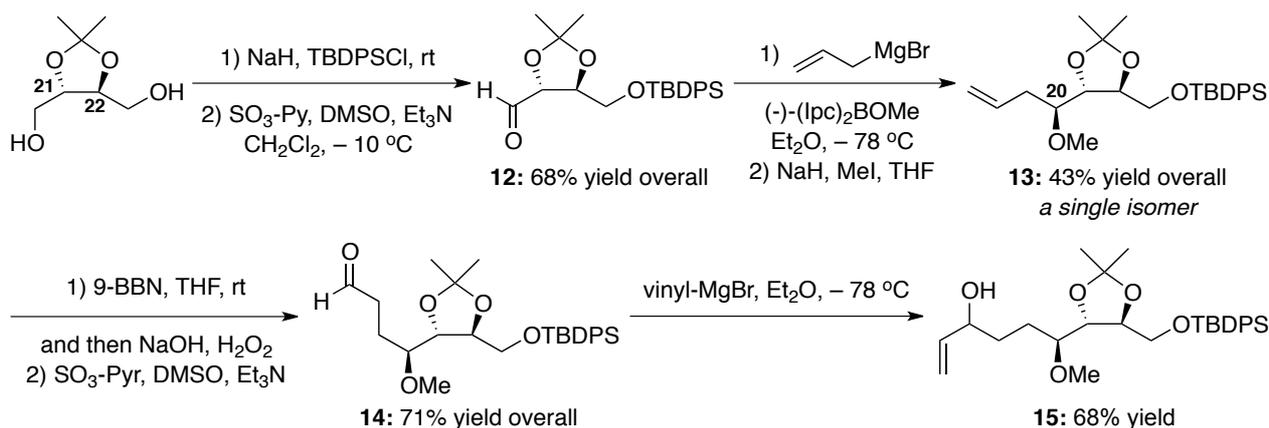
**Scheme 2.** A Facile Assembly of the Pyranyl A-Ring **11**.

A directed-reductive ring-opening of the epoxide proceeded regioselectively and a subsequent diol protection using 2,2-dimethoxy propane gave acetonide **7**. With the optically enriched acetonide **7** in hand, we proceeded to complete the pyran synthesis. Debencylation followed by Doering–Parikh oxidation<sup>11b</sup> gave aldehyde **8**. Wittig olefination<sup>11b</sup> and hydrolytic removal of the acetonide group occurred concomitantly with the pyran formation yielded pyran **10** in 87% overall yield as a single diastereomer. The relative *syn* stereochemical relationship at C3 and C7 was confirmed through NOE. Although kinetic control has been noted,<sup>12</sup> the high level of stereoselectivity is likely a result of thermodynamically controlled *O*-1,4-addition, leading to complete chirality transfer from C3 to C7. Subsequent Piv-protection of C1-OH in **10** gave methyl ketone **11**, which would then complete our asymmetric synthesis of A-ring, and set up the critical C10 and C11 connection via an *anti*-aldol [Scheme 1].

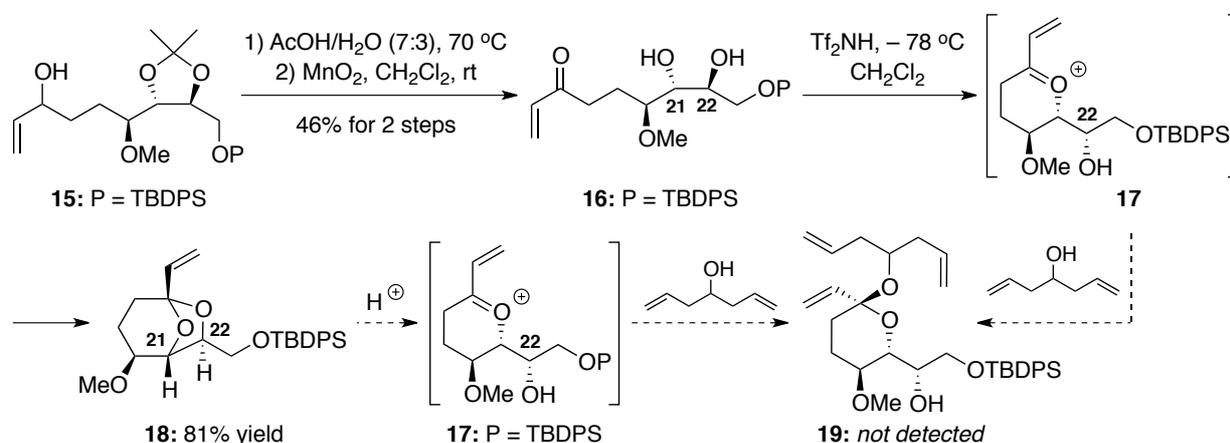
### 3. BC-Ring Construction.

**Synthesis of the Key Cyclic Acetal.** Unlike the A-ring synthesis that featured asymmetric catalysis, to prepare BC-ring, we elected the Hanessian's chiron-pool concept and commenced with commercially available (+)-2,3-(*O*)-*iso*-propylidene-L-threitol,<sup>10b,13a</sup> thereby borrowing the desired C21 and C22 stereochemistry. Mono-protection of the hydroxyl group in (+)-2,3-(*O*)-*iso*-propylidene-L-threitol with TBDPSCl followed by Doering–Parikh oxidation<sup>11b</sup> gave aldehyde **12** in 68% yield [Scheme 3]. A

five-carbon chain extension of aldehyde **12** was accomplished with a five-step sequence: (i) Brown's asymmetric allylation to set up the C20 stereochemistry in a reagent controlled manner,<sup>11a</sup> (ii) standard *O*-methylation to give methyl ether **13**; (iii) classical 9-BBN hydroboration of **13**; (iv) Doering–Parikh oxidation,<sup>11b</sup> or modified-Moffat protocol, to afford aldehyde **14**; and (v) vinyl Grignard addition to yield allylic alcohol **15**. Stereochemistry at C20 was confirmed using the classical Mosher ester analysis<sup>11c,d</sup> after the Brown's asymmetric allylation step using **12**.



**Scheme 3.** Synthesis of Aldehyde **14** and Allyl Alcohol **15**.

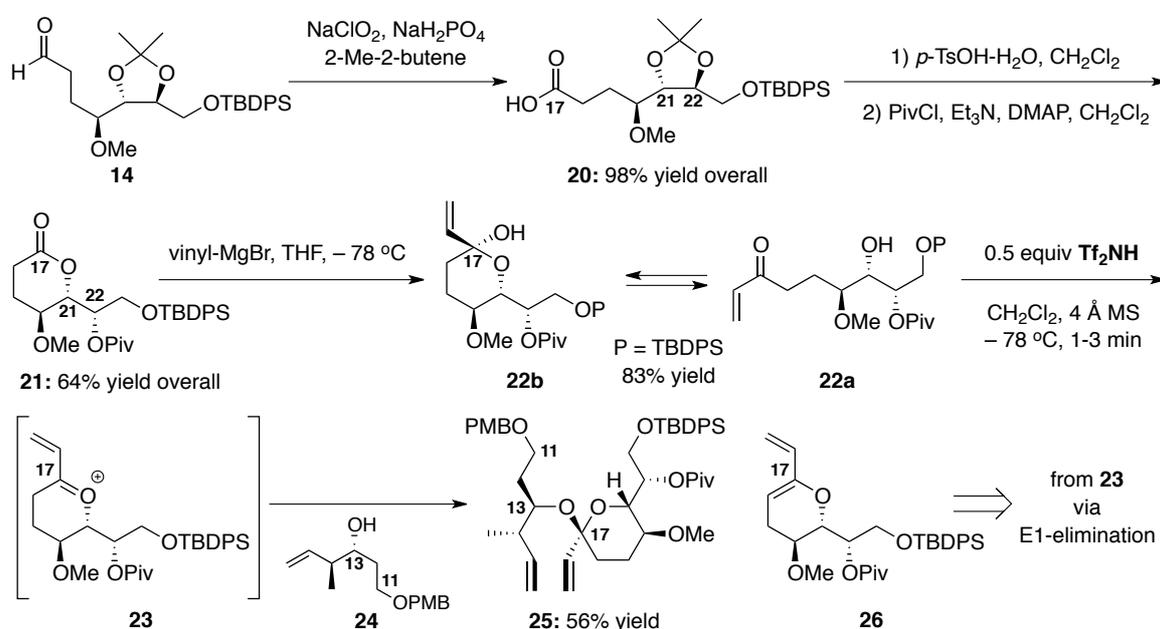


**Scheme 4.** Failed Attempt on Cyclic Acetal Construction from Allyl Alcohol **15**.

At this juncture, we recognizing the potential of a more facile route to the key vinyl cyclic acetal, and thus, we proceeded to remove the acetonide group in allylic alcohol **15**, and a subsequent  $\text{MnO}_2$  oxidation gave enone **16** that contains the free diol at C21 and C22 [**Scheme 4**]. However, attempts to access cyclic acetal **19** from **16** through oxocarbenium ion **17** via acid promoted condensation with 1,6-heptadiene-4-ol failed. From these attempts,<sup>14</sup> we were only able to observe and/or isolate bicyclic acetal **18** in 81% yield when using  $\text{Tf}_2\text{NH}$ .<sup>15-17</sup> The formation of **18** is clearly a result of trapping of vinyl oxocarbenium ion **17**

by the free C22-OH in a facile intramolecular manner. Given that **19** and **18** are both derived through **17** in a reversible manner, we tried but again failed at longer reaction time and higher temperatures to force **18** equilibrating toward **19**.

After much experimentation, we ultimately succeeded in synthesizing the key vinyl cyclic acetal **25** via the route shown in **Scheme 5**. Lindgren oxidation<sup>18</sup> of aldehyde **14** afforded acid **20**. Removal of the acetonide group led to a selective lactone formation involving only C21-OH, and subsequent capping of C22-OH with PivCl gave lactone **21** in 64% overall yield. The ensuing addition of vinyl magnesium bromide followed by treatment of the resulting lactol mixture **22a/b** with  $\text{Tf}_2\text{NH}$ <sup>17</sup> in the presence of alcohol **24**<sup>19a,b</sup> afforded the desired vinyl cyclic acetal **25** in 56% yield, albeit still accompanied with the elimination product **26** that was not easily separable.

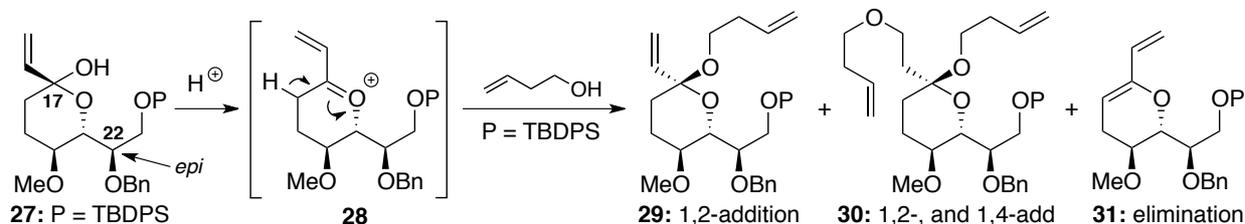


**Scheme 5.** A Successful Route to Cyclic Acetal **25** from Aldehyde **14**.

**Exploring Conditions for the Cyclic Acetal Formation.** Although we have developed our own unique protocols for the acetal formation using simple pyranyl systems<sup>7,8</sup> and that there are ample reports for more robust carbohydrate systems,<sup>20</sup> it is worthy to note that this particular cyclic acetal formation took some efforts to investigate. As summarized in **Table 1**, formation of cyclic acetal using lactol **27** [*epi* at C22], which contains the anomeric vinyl group at C17, proved to be challenging. A range of acids as well as solvents and temperatures were screened. While most frequently used Lewis acids and Brønsted acids in anomeric substitutions<sup>20</sup> led to the over addition product **30**,  $\text{Tf}_2\text{NH}$  [entry 5] proved to be an excellent Brønsted acid at  $-78^\circ\text{C}$ , leading to **29** as the sole product in 89% yield as a single diastereomer with the oxo-butenyl group being axial.<sup>13b</sup> The ability of  $\text{Tf}_2\text{NH}$  in leading to the desired outcome in an array of

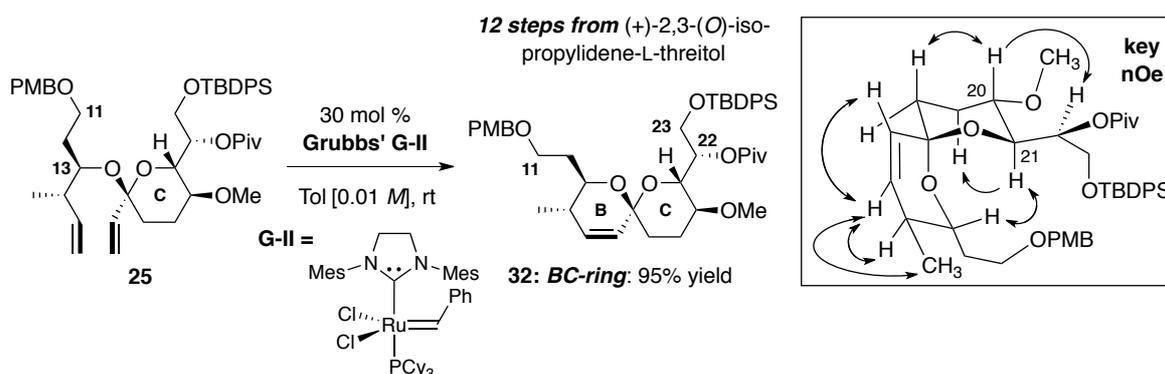
reaction pathways have been well noted,<sup>15,16</sup> albeit poorly understood other than the fact that represents one softest anionic complex.<sup>17</sup>

**Table 1.** A Comparison of Acid Promoters in the Cyclic Acetal Formation.



| entry | acid                               | equiv   | solvent                         | temp      | time    | product         |
|-------|------------------------------------|---------|---------------------------------|-----------|---------|-----------------|
| 1     | BF <sub>3</sub> -Et <sub>2</sub> O | 1.0     | CH <sub>2</sub> Cl <sub>2</sub> | -78 °C    | 100 min | <b>30</b> [82%] |
| 2     | TMSOTf                             | 1.0     | CH <sub>2</sub> Cl <sub>2</sub> | 0         | 15 min  | <b>30</b> [88%] |
| 3     | K-10                               | 1.2-2.5 | benzene                         | rt-reflux | 5-15 h  | <b>30</b> [89%] |
| 4     | CSA or PPTS                        | 0.5-1.0 | CH <sub>2</sub> Cl <sub>2</sub> | rt-reflux | 1-15 h  | No rxn          |
| 5     | Tf <sub>2</sub> NH                 | 1.0     | CH <sub>2</sub> Cl <sub>2</sub> | -78       | 5 min   | <b>29</b> [89%] |

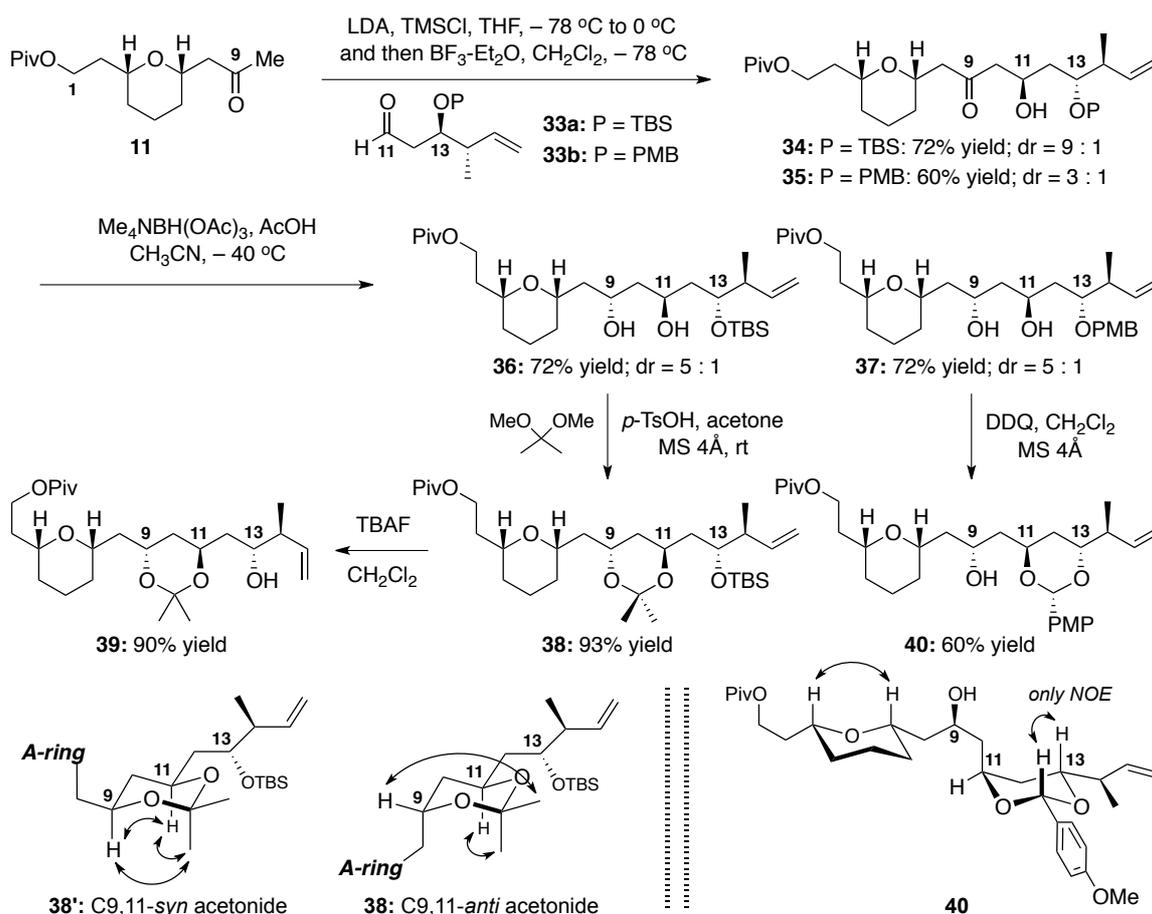
**Cyclic Acetal-Tethered RCM.** At last, the key ring-closing metathesis [RCM] of vinyl cyclic acetal **25** led to BC-ring **32** in 95% yield using Grubbs' Generation-II Ru-catalyst [Scheme 6].<sup>21</sup> It is noteworthy that NOE experiments of **32** revealed its the desired relative stereochemistry, which closely resemble those reported for the same region in spirastrellolide A.<sup>1</sup> More importantly, the current synthesis of the C11-C23 fragment **32** took only 12 steps from (+)-2,3-(*O*)-*iso*-propylidene-L-threitol.<sup>13a</sup> This chiron approach is much shorter and synthetically more practical comparing to the previous 18-step route from D-glucose, which also led to *epi*-stereochemistry at C22.<sup>13b</sup> We should point out that our synthetic study represented the first in the area of spirastrellolide synthesis when we pursued the 18-step C22-*epi*-synthesis, and the correct assignment had not been reported by Roberge and Andersen *et al.* Our intent at the time was to simply showcase our strategy.



**Scheme 6.** Synthesis of BC-Ring **32** via Cyclic Acetal Tethered RCM.

#### 4. Attempted Convergent Approach to C1-C23.

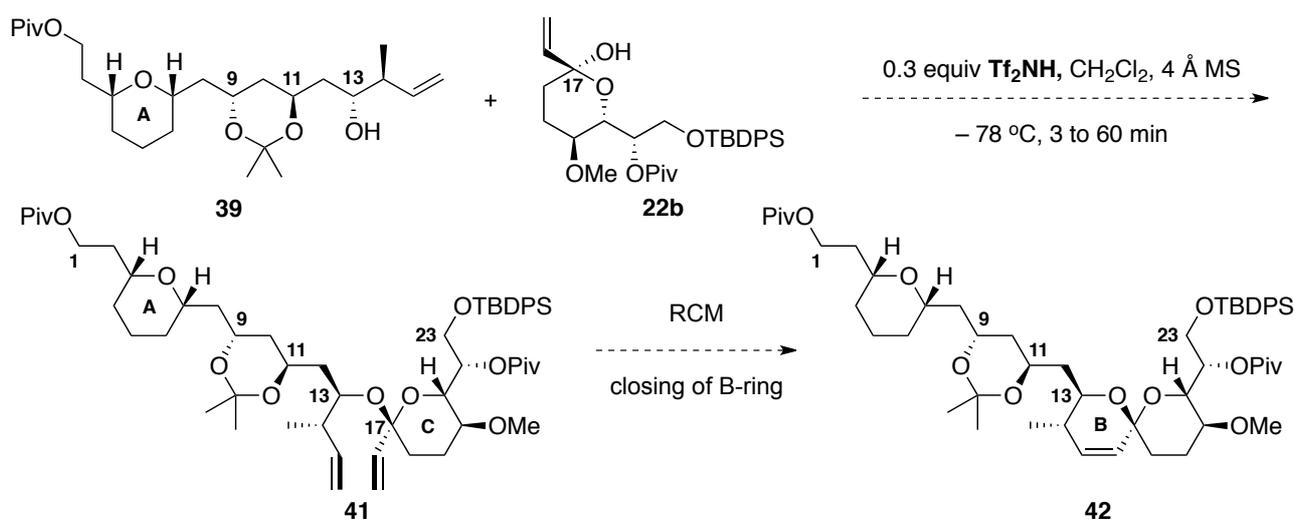
To examine the concept of connecting A-ring and BC-ring at C10 and C11 through a diastereoselective aldol addition, and to explore a possibly more convergent route, we prepared aldehyde **33a**<sup>19a,b</sup> and **33b**.<sup>19c</sup> By employing Mukaiyama's conditions,<sup>22</sup> methyl ketone **11** was first converted to its respective TMS-enol ether using LDA and TMSCl, and the resulting TMS-enol ether was added to aldehyde **33a** [or **33b**] followed by the addition of a stoichiometric amount of  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  at  $-78^\circ\text{C}$  to give the aldol product **34** in 72% yield with a diastereomeric ratio of 9:1 [Scheme 7]. The major isomer was initially assigned as the desired C11,13-*anti* product based on Evans' non-chelation 1,3-asymmetric induction model.<sup>23</sup> By using aldehyde **33b** under the same conditions, the yield for the respective aldol product **35** was comparable but the *dr* was only 3:1. It is noteworthy that following Evans boron enolate conditions<sup>24</sup> via di-*n*-butylboron enolate derived from **11** and aldehyde **33a** in  $\text{CH}_2\text{Cl}_2$  at  $-78^\circ\text{C}$  afforded **34** with 29% yield but in  $\sim 1:1$  ratio.



Scheme 7. Synthesis of Alcohol **39** and PMP-C11,13-*Anti*-Acetal **40**.

A directed reduction of hydroxyl ketone **34** and **35** using  $\text{Me}_4\text{NBH}(\text{OAc})_3$  led to diols **36** and **37**, respectively. The diastereoselectivity here is modest with the *dr* being 5:1 for both diols **36** and **37**, the

major isomer indeed favored C9 and C11 *anti* in relative stereochemistry. This *anti*-selectivity was unambiguously confirmed through NOE experiments using C9,11-*anti* acetonide **38** that could be prepared from C13-TBS-protected C9,11-*anti*-diol **36**. To be ascertain, C9,11-*syn* acetonide **38'** was also prepared in an analogous sequence to obtain a contrasting NOE outcome [Scheme 7].<sup>25</sup> On the other hand, by using C13-PMB-protected C9,11-*anti*-diol **37**, we were able to finally unambiguously assign the C11,13-*anti* relative stereochemistry through NOE of the corresponding PMP acetal **40**, which could be accessed via DDQ oxidation of **37**.



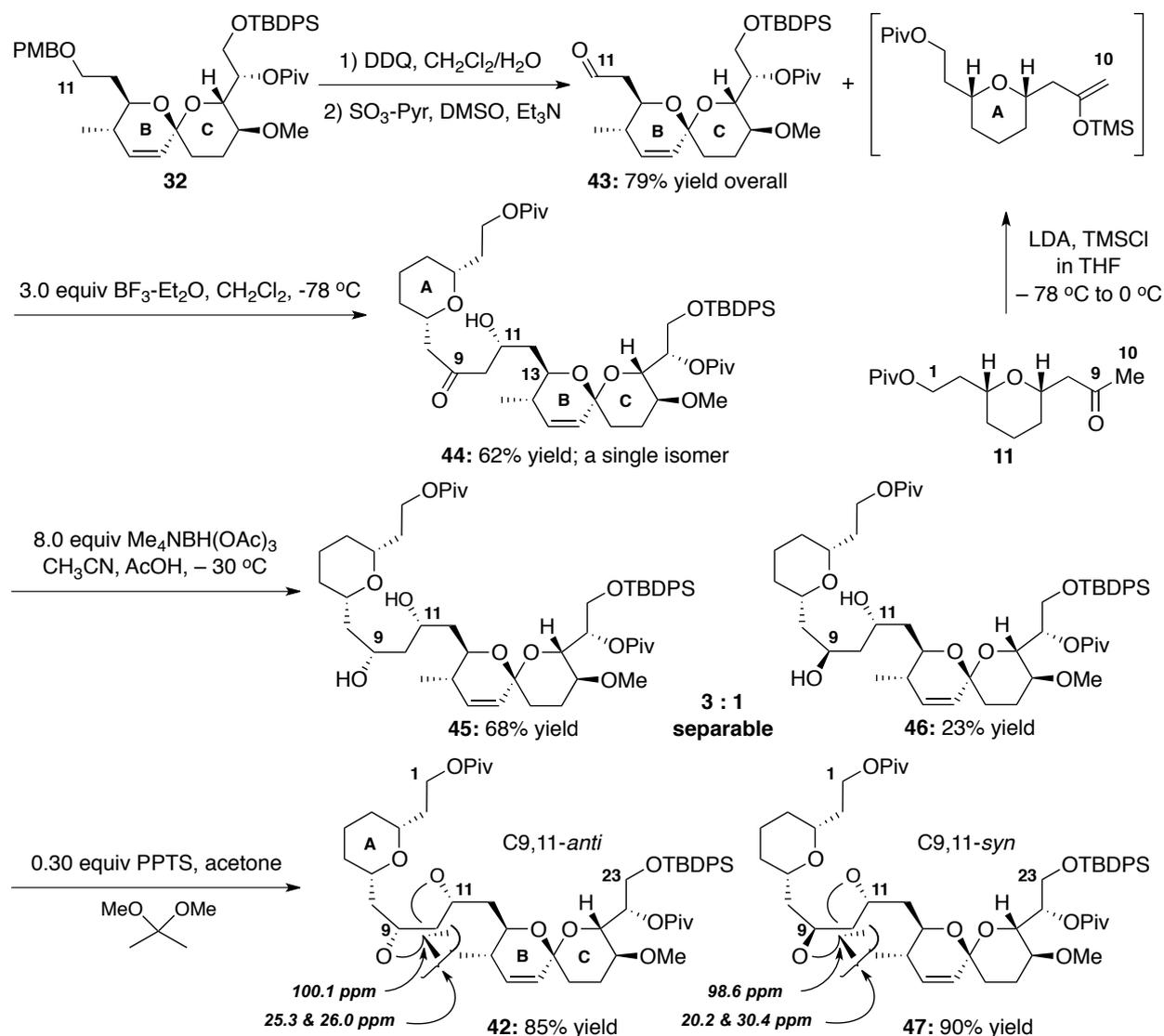
**Scheme 8.** A Failed Convergent Approach to ABC-Ring **42** via Cyclic Acetal Tethered RCM of **41**.

While syntheses of C9,11-*anti* acetonide **38** was quintessential in confirming key stereochemical assignments, we were also attempting to explore a more convergent route to the Southern Half by first connecting the A-ring and C-ring through cyclic acetal formation, linking first together C13-OH with C17. However, that possibility did not work well. As shown in **Scheme 8**, attempted synthesis of vinyl cyclic acetal **41** using alcohol **39** [via desilylation of **38**] and lactol **22b** was not successful under standard conditions as well as other conditions. Thus, this thwarted our efforts to assembly the entire ABC tricycle **42** through closing of the B-ring at the very end via the cyclic acetal tethered RCM strategy.

## 5. Completing the Southern Macrocylic Half.

To complete the synthesis of the *Southern Half* of the macrocycle G-ring, BC-ring **32** was transformed to aldehyde **43** via removal of the PMB protecting group and  $\text{SO}_3\text{-Pyr/DMSO}$  oxidation [Scheme 9]. To connect the C10-C11 bond, aldehyde **43** was subjected to the same Mukaiyama aldol conditions<sup>21</sup> employing TMS-enol ether derived *in situ* from methyl ketone **11**. It was once again important here that TMS-enol ether was first added to aldehyde **43** prior to the addition of a stoichiometric amount of

$\text{BF}_3\text{-Et}_2\text{O}$  at  $-78^\circ\text{C}$ . The aldol product **44** was obtained as a single isomer in 62% yield with C11-C13 relative stereochemistry assigned as *anti* based on the study described above in the preparation of **40**.



**Scheme 9.** A Successful Assembly of the Southern Macrocyclic Half.

This assignment is again also consistent with Evans non-chelation 1,3-asymmetric induction dipole directed model.<sup>22</sup> Directed reduction led to diols **45** and **46** with a 3:1 ratio in favor of the C9,11-*anti* isomer. Although the selectivity was low, both isomers would serve a real purpose. Acetonide formation of both *anti* and *syn* diol isomers **45** and **46** gave the desired C9,11-*anti* acetonide **42**, and C9,11-*syn* acetonide **47**, respectively. With both *syn* and *anti* isomers in hand, we could unambiguously assign the C9 and C11 relative stereochemistry in both **42** and **47** through the Rychnovsky-Evans C-13 analysis,<sup>26,27</sup> thereby completing our endeavor in the assembly of the entire Southern macrocyclic Half of (+)-spirastrellolide A.

## CONCLUSION

We have described here details of our efforts toward the construction of both A-ring and BC-ring of (+)-spirastrellolide A. The A-ring synthesis constituted a facile 12-step linear sequence starting from 1,5-pentanediol, and the BC-ring synthesis would ultimately showcase a cyclic acetal-tethered RCM strategy that was developed in our lab for *de novo* synthesis of spiroketals. Assembly of the entire Southern Half of this unique macrocycle would also require 1,3-*anti*-Mukaiyama aldol addition for connecting A-ring and BC-ring specifically between C10 and C11, thereby culminating a successful and practical 17-step approach linearly from (+)-2,3-(*O*)-iso-propylidene-L-threitol. An attempt toward an even more convergent synthesis of the Southern Half was also carried out. However, our efforts failed in connecting A-ring and C-ring through an acetal formation that would have linked together the free C13-OH with C17 at the spiro-BC-ring junction. This cyclic acetal intermediate was to be subjected to our cyclic acetal-tethered RCM strategy to close the B-ring. Nevertheless, the current 17-step approach should prove to be highly practical for an ultimate total synthesis of (+)-spirastrellolide A.

## EXPERIMENTAL

### Synthesis of Allyl Alcohol 5.

To a slurry of NaH (9.22 g, 230 mmol) in anhyd THF was added dropwise 1,5-pentanediol at 0 °C. After stirring this slurry for 30 min, benzyl bromide (25.6 mL, 211.2 mmol) was added. The reaction mixture was refluxed for 12 h before being quenched with dropwise addition of H<sub>2</sub>O. The organic phase was extracted with EtOAc, washed with sat aq NaCl, and dried over Na<sub>2</sub>SO<sub>4</sub>. The solvent was concentrated under reduced pressure and the crude residue was purified by flash silica gel column chromatography [33% EtOAc in hexane] to provide mono-benzylated diol as colorless oil (21.0 g) in 56% yield.  $R_f = 0.30$  [33% EtOAc in hexane]; <sup>1</sup>H NMR (500MHz, CDCl<sub>3</sub>) δ 1.41-1.48 (m, 2H), 1.58 (tt,  $J = 6.5, 6.5$  Hz, 2H), 1.64 (tt,  $J = 7.0, 7.0$  Hz, 2H), 2.02 (brs, 1H), 3.48 (t,  $J = 6.5$  Hz, 2H), 3.62 (t,  $J = 7.0$  Hz, 2H), 4.51 (s, 2H), 7.26-7.36 (m, 5H); <sup>13</sup>C NMR (125MHz, CDCl<sub>3</sub>) δ 22.7, 29.7, 32.8, 63.1, 70.6, 73.2, 127.8, 127.9, 128.7, 138.8; IR (neat) cm<sup>-1</sup> 3393br, 3031s, 2936s, 2861s, 2360s, 2342s, 1455s, 1363s, 1098m; mass spectrum (APCI):  $m/z$  (% relative intensity) 195 (M+H)<sup>+</sup> (75), 191 (8), 184 (9), 168 (10), 145 (19), 139 (15), 117 (30), 101 (100), 100 (51).

To a solution of oxalyl chloride (10.5 mL, 118 mmol) in CH<sub>2</sub>Cl<sub>2</sub> was added a solution of DMSO (17 mL, 238.0 mmol) in CH<sub>2</sub>Cl<sub>2</sub> at -78 °C. After the addition, the mixture was stirred for 30 min. before the above mono-benzylated diol (21.0 g, 108.0 mmol) was added to dropwise over 20 min period. The mixture was stirred for an additional 1 h at -78 °C before Et<sub>3</sub>N (80.0 mL, 570.0 mmol) was added through a syringe, and the mixture was slowly warmed up to rt. Subsequently, the reaction mixture was

washed with H<sub>2</sub>O and sat aq NH<sub>4</sub>Cl. The organic phase was extracted with EtOAc, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The crude residue was purified by flash silica gel column chromatography [20% EtOAc in hexane] to aldehyde **4** as yellow oil (18.0 g) in 87% yield.  $R_f = 0.33$  [20% EtOAc in hexane]; <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>) δ 1.59-1.67 (m, 2H), 1.68-1.76 (m, 2H), 2.42 (td,  $J = 2.0, 7.2$  Hz, 2H), 3.47 (t,  $J = 6.4$  Hz, 2H), 4.51 (s, 2H), 7.24-7.35 (m, 5H), 9.72 (t,  $J = 2.0$  Hz, 1H); <sup>13</sup>C NMR (125MHz, CDCl<sub>3</sub>) δ 19.2, 29.4, 43.8, 70.0, 73.2, 127.8, 127.9, 128.6, 138.7, 202.7; IR (neat) cm<sup>-1</sup> 2936s, 2860s, 1720s; mass spectrum (APCI):  $m/z$  (% relative intensity) 215 (M+Na)<sup>+</sup> (11), 171 (8), 145 (41), 121 (20), 115 (55), 101 (100).

To a solution of aldehyde **4** (18.0 g, 94.0 mmol) in CH<sub>2</sub>Cl<sub>2</sub> was added Wittig reagent Ph<sub>3</sub>P=CHCO<sub>2</sub>Et (50.0 g, 141.0 mmol). After stirring at rt for 5 h, excess of CH<sub>2</sub>Cl<sub>2</sub> was evaporated under reduced pressure and the residue was diluted with a small amount of hexane. The slurry was filtered through a pad of Celite<sup>TM</sup> to remove triphenylphosphine oxide. The filtrate was concentrated under reduced pressure. The crude residue was purified by flash silica gel column chromatography [10% EtOAc in hexane] to provide Wittig olefination product as colorless oil (24.0 g) in 98% yield and ≥25:1 E:Z selectivity.  $R_f = 0.30$  [10% EtOAc in hexane]; <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>) δ 1.27 (t,  $J = 7.0$  Hz, 3H), 1.51-1.59 (m, 2H), 1.60-1.67 (m, 2H), 2.20 (ddt,  $J = 2.0, 7.2, 14.4$  Hz, 2H), 3.46 (t,  $J = 6.0$  Hz, 2H), 4.17 (q,  $J = 7.0$  Hz, 2H), 4.50 (s, 2H), 5.81 (dt,  $J = 1.6, 15.6$  Hz, 1H), 6.95 (dt,  $J = 6.8, 15.6$  Hz, 1H), 7.24-7.36 (m, 5H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 14.6, 25.0, 29.5, 32.2, 60.4, 70.2, 73.2, 121.8, 127.8, 127.9, 128.7, 138.8, 149.2, 167.0; IR (neat) cm<sup>-1</sup> 2980m, 2937s, 2859s, 2360br, 1719s, 1654s; mass spectrum (APCI):  $m/z$  (% relative intensity) 263 (M+H)<sup>+</sup> (7), 218(15), 217(100), 200(7), 199(52), 175(7), 171(10), 157(10).

To a solution of the above Wittig olefination product (24.0 g, 92.0 mmol) in THF was added DIBAL-H (1.0 M in toluene, 239.0 mL, 230.0 mmol) at -78 °C. The mixture was gradually warmed up to -50 °C and stirred for an additional 1.5 h before being quenched with aq HCl (1.0 M, 150 mL). The reaction mixture was extracted with EtOAc, and combined organic layers was dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The crude residue was purified with flash silica gel column chromatography [33% EtOAc in hexane] to provide allyl alcohol **5** (20.0 g) in 98 % yield.  $R_f = 0.35$  [33% EtOAc in hexane]; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 1.44-1.51 (m, 2H), 1.60-1.67 (m, 2H), 1.80 (s, 1H), 2.04-2.09 (q,  $J = 7.2$  Hz, 2H), 3.47 (t,  $J = 6.4$  Hz, 2H), 4.08 (d,  $J = 4.0$  Hz, 2H), 4.50 (s, 2H), 5.60-5.73 (m, 2H), 7.25-7.37 (m, 5H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 26.0, 29.5, 32.2, 63.8, 70.4, 73.1, 127.8, 127.9, 128.6, 129.6, 132.9, 138.8; IR (neat) cm<sup>-1</sup> 3393brs, 2934s, 2858s, 2361s, 1455s, 1363s; mass spectrum (ESI):  $m/z$  (% relative intensity) 243 (M+Na)<sup>+</sup> (100),

### Sharpless Asymmetric Epoxidation of Allyl Alcohol 5.

To a suspension of dried and pulverized Molecular Sieve 4Å in anhyd CH<sub>2</sub>Cl<sub>2</sub> (6 mL) was added D-(-)-di-isopropyl tartrate (0.051 mL, 0.24 mmol) followed by slow addition of Ti(O*i*-Pr)<sub>4</sub> (0.048 mL, 0.16 mmol) at -20 °C. After which the above allyl alcohol **5** (176.6 mg, 0.802 mmol) was added and the mixture was stirred for 5 min. before TBHP (3.4 M, 0.71 mL, 2.40 mmol) was added slowly and the resulting solution was stirred at -20 °C for 1.5 h. The reaction was then quenched with aq NaOH (1.0 N) and was stirred for an additional 1 h at 0 °C before it was warmed up to rt. The reaction mixture was filtered through a pad of Celite<sup>TM</sup>. The organic phase was extracted with CH<sub>2</sub>Cl<sub>2</sub>, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The crude residue was purified by flash silica gel column chromatography [50% EtOAc in hexane] to provide the desired epoxy alcohol **6** (164.4 mg) in 87% yield and 96% *ee*. The enantiomeric excess was determined using chiral HPLC. The retention times of the two enantiomers were 25.789 (S) min and 27.861 (R) min respectively.  $R_f = 0.10$  [30% EtOAc in hexane];  $[\alpha]_D^{23} = 36.6$  [ $c = 0.89$ , CH<sub>2</sub>Cl<sub>2</sub>]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  1.42-1.60 (m, 4H), 1.60-1.69 (m, 2H), 2.20 (brt,  $J = 5.0$  Hz, 1H), 2.91 (dt,  $J = 2.5, 4.0$  Hz, 1H), 2.95 (dt,  $J = 2.5, 5.5$  Hz, 1H), 3.49 (t,  $J = 6.5$  Hz, 2H), 3.62 (ddd,  $J = 4.5, 7.0, 11.5$  Hz, 1H), 3.89 (ddd,  $J = 2.5, 5.5, 13.0$  Hz, 1H), 4.51 (s, 2H), 7.28-7.30 (m, 1H), 7.30-7.35 (m, 4H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  22.7, 29.5, 31.4, 56.0, 58.7, 61.9, 70.1, 72.9, 127.6, 127.7, 128.4, 138.5; IR (neat) cm<sup>-1</sup> 3434brs, 3030m, 2936s, 2861m, 1100s; mass spectrum (APCI):  $m/z$  (% relative intensity) 237.2 (M+H)<sup>+</sup> (100);  $m/z$  calcd for C<sub>14</sub>H<sub>20</sub>O<sub>3</sub>Na<sup>+</sup> 259.1310, found 259.1305.

### Preparation of Acetonide 7.

To a solution of epoxy alcohol **6** (1.80 g, 7.50 mmol) in THF (10 mL) was added Red-Al (8.0 mL, 26.3 mmol) dropwise at -10 °C. The mixture was warmed up to rt and stirred for an additional 3 h. Subsequently, sat aq NH<sub>4</sub>Cl (20 mL) was added and the resulting mixture was stirred for another 1 h. The reaction mixture was extracted with Et<sub>2</sub>O, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The crude residue was purified by flash silica gel column chromatography [91% EtOAc in hexane] to the desired diol (1.76 g) in 98% yield.  $R_f = 0.33$  [91% EtOAc in hexane]; <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>)  $\delta$  1.39-1.57 (m, 4H), 1.59-2.04 (m, 4H), 2.81 (brs, 1H), 2.91 (brs, 1H), 3.50 (t,  $J = 6.4$  Hz, 2H), 3.78-3.85 (m, 3H), 4.50 (s, 2H), 7.25-7.37 (m, 5H); <sup>13</sup>C NMR (125MHz, CDCl<sub>3</sub>)  $\delta$  22.5, 29.9, 37.8, 38.5, 62.1, 70.6, 72.3, 73.2, 127.8, 128.0, 128.7, 138.8; IR (neat) cm<sup>-1</sup> 3366brs, 2933s, 2858s, 2360s, 2341s; mass spectrum (ACPI):  $m/z$  (% relative intensity) 239 (M+H)<sup>+</sup> (7), 221(17), 203(24), 185(12), 143(13), 131(16), 129(32), 117(7), 113(100), 111(9).

To a solution of the above diol (161.0 mg, 0.68 mmol) in acetone was added molecular sieve 4Å and

dimethoxy-propane (0.25 mL, 2.03 mmol) followed by *p*-TsOH·H<sub>2</sub>O (12.8 mg, 0.068 mmol). The mixture was stirred for 12 h before it was quenched with sat aq NaHCO<sub>3</sub> (10 mL). The reaction mixture was extracted with EtOAc, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The crude residue was purified by flash silica gel column chromatography [17% EtOAc in hexane] to provide acetone 7 (170.9 mg) in 91% yield. *R*<sub>f</sub> = 0.35 [17% EtOAc in hexane]; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 1.38 (s, 3H), 1.38-1.43 (m, 2H), 1.44 (s, 3H), 1.46-1.58 (m, 4H), 1.63 (tt, *J* = 6.8, 6.8 Hz, 2H), 3.47 (t, *J* = 6.8 Hz, 2H), 3.82 (ddd, *J* = 1.6, 5.2, 12.0 Hz, 2H), 3.95 (td, *J* = 2.8, 12.4 Hz, 1H), 4.54 (s, 2H), 7.26-7.37 (m, 5H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 19.5, 21.8, 29.9, 30.3, 31.5, 36.5, 60.3, 69.0, 70.5, 73.1, 98.4, 127.7, 127.9, 128.6, 138.9; IR (neat) cm<sup>-1</sup> 2992s, 2940s, 2862s, 2360s; mass spectrum (ESI): *m/z* (% relative intensity) 301.7 (M+Na)<sup>+</sup> (100); *m/z* calcd for C<sub>17</sub>H<sub>26</sub>O<sub>3</sub>Na<sup>+</sup> 301.1780, found 301.1774.

### Synthesis of Aldehyde 8

A suspension of acetone 7 (587.5 mg, 2.10 mmol) and 10% Pd/C (45.0 mg) in EtOAc was stirred under a balloon of hydrogen at rt for 1 h. The reaction mixture was filtered through a pad of Celite<sup>TM</sup> and the filtrate was concentrated under reduced pressure. The crude oil was used for the next step without further purification. *R*<sub>f</sub> = 0.50 [83% EtOAc in hexane]; <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>) δ 1.33 (s, 3H), 1.33-1.40 (m, 2H), 1.40 (s, 3H), 1.42-1.58 (m, 6H), 2.22 (brs, 1H), 3.58 (t, *J* = 6.4 Hz, 2H), 3.80-3.88 (m, 2H), 3.93-3.99 (td, *J* = 2.8, 12.0 Hz, 1H); <sup>13</sup>C NMR (125MHz, CDCl<sub>3</sub>) δ 19.4, 21.4, 30.2, 31.5, 32.7, 36.3, 60.2, 62.7, 69.1, 98.4.

To a solution of the above crude alcohol in CH<sub>2</sub>Cl<sub>2</sub>/DMSO (2.5 mL, 10.6 mmol) were added Et<sub>3</sub>N (1.60 mL, 11.3 mmol) and SO<sub>3</sub>·pyridine complex (1.4 g, 8.5 mmol) at 0 °C. The resulting mixture was stirred at 0 °C for 1.5 h before it was quenched with H<sub>2</sub>O and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with sat aq NaCl, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The crude residue was purified with flash silica gel column chromatography [33% EtOAc in hexane] to provide aldehyde 8 (203.4 mg) in 52% yield. *R*<sub>f</sub> = 0.45 [33% EtOAc in hexane]; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 1.38 (s, 3H), 1.39-1.44 (m, 1H), 1.45 (s, 3H), 1.47-1.83 (m, 5H), 2.46 (td, *J* = 1.6, 6.4 Hz, 2H), 3.81-3.88 (m, 2H), 3.93-3.99 (td, *J* = 2.8, 12.0 Hz, 1H), 9.77 (t, *J* = 1.6 Hz, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 18.0, 19.5, 30.2, 31.5, 36.0, 44.0, 60.2, 68.8, 98.5, 202.8; IR (neat) cm<sup>-1</sup> 3300m, 2993s, 2938s, 2869s, 1718s.

### Preparation of A-Ring 11.

To a solution of aldehyde 8 (23.0 mg, 0.12 mmol) in THF (1 mL) was added Wittig reagent Ph<sub>3</sub>P=CHCOMe (99.3 mg, 0.31 mmol) at rt and the mixture was refluxed for 5 h. The solution was then

cooled down to rt and concentrated under reduced pressure. The crude residue was purified with flash column chromatography [33% EtOAc in hexane] to provide the desired *E*-enone (26.4 mg, 0.12 mmol) in 95% yield.  $R_f = 0.37$  [33% EtOAc in hexane];  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  1.38 (s, 3H), 1.39-1.43 (m, 2H), 1.45 (s, 3H), 1.46-1.79 (m, 6H), 2.25 (s, 3H), 3.81-3.88 (m, 2H), 3.96 (dt,  $J = 2.8, 12.0$  Hz, 1H), 6.08 (dt,  $J = 1.6, 16.0$  Hz, 1H), 6.80 (dt,  $J = 7.2, 16.0$  Hz, 1H);  $^{13}\text{C NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  19.5, 23.8, 27.1, 30.3, 31.5, 32.6, 36.2, 60.2, 68.8, 98.5, 131.8, 148.3, 198.9; IR (neat)  $\text{cm}^{-1}$  3750m, 3675m, 3649m, 3629m, 2992s, 2940s, 2865s, 2340s, 1698s, 1673s, 1626s;

To a solution of the above *E*-enone (81.2 mg, 0.36 mmol) in  $\text{CH}_2\text{Cl}_2$  was added *p*-TsOH $\cdot$ H $_2$ O (20.8 mg, 0.11 mmol) at 0 °C. The mixture was stirred for 1 h before it was quenched with sat aq  $\text{NaHCO}_3$ . The reaction mixture was extracted with  $\text{CH}_2\text{Cl}_2$ , and the combined organic layers were dried over  $\text{Na}_2\text{SO}_4$  and concentrated under reduced pressure. The crude residue was purified by flash silica gel column chromatography [67% EtOAc in hexane] to provide pyran **10** (60.3 mg) as a colorless oil in 91% yield.  $R_f = 0.33$  [67% EtOAc in hexane];  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  1.17-1.35 (m, 2H), 1.51-1.65 (m, 4H), 1.66-1.76 (m, 1H), 1.81-1.87 (m, 2H), 2.16 (s, 3H), 2.47 (dd,  $J = 4.8, 16.0$  Hz, 1H), 2.57 (brs, 1H), 2.67 (dd,  $J = 7.6, 16.0$  Hz, 1H), 3.60 (dddd,  $J = 2.0, 3.6, 8.4, 12.8$  Hz, 1H), 3.76 (dt,  $J = 5.6, 5.6$  Hz, 2H), 3.83 (dddd,  $J = 2.0, 4.8, 8.0, 12.8$  Hz, 1H);  $^{13}\text{C NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  23.5, 31.1, 31.3, 31.4, 38.2, 50.3, 61.5, 74.2, 78.5, 207.5; IR (neat)  $\text{cm}^{-1}$  3852m, 3675m, 3649m, 3629m, 3376brs, 2934s, 2860s, 2340s, 1716s, 1670s; mass spectrum (APCI):  $m/z$  (% relative intensity) 187 ( $\text{M}+\text{H}$ ) $^+$  (100), 169 (24), 151(27), 129(84), 111(18); mass spectrum (ESI):  $m/z$  calcd for  $\text{C}_{10}\text{H}_{18}\text{O}_3\text{Na}^+$  209.1154, found 209.1148.

To a solution of pyran **10** (60.3 mg, 0.32 mmol) in  $\text{CH}_2\text{Cl}_2$  was added pyridine (0.11 mL, 1.30 mmol). After stirring at rt for 0.5 h and pivaloyl chloride (0.08 ml, 0.65 mmol) was added. The mixture was stirred for an additional 1 h before it was quenched with  $\text{H}_2\text{O}$ . The organic phase was extracted with  $\text{CH}_2\text{Cl}_2$ , dried over  $\text{Na}_2\text{SO}_4$ , and concentrated under reduced pressure. The crude residue was purified with flash silica gel column chromatography [25% EtOAc in hexane] to provide A-ring **11** or the C1-C11 fragment (76.2 mg) as yellowish oil in 87% yield.  $R_f = 0.40$  [25% EtOAc in hexane];  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  1.19 (s, 9H), 1.21 (m, 2H), 1.53-1.63 (m, 3H), 1.75 (dd,  $J = 7.0, 13.0$  Hz, 2H), 1.82-1.85 (m, 1H), 2.18 (s, 3H), 2.41 (dd,  $J = 5.0, 15.0$  Hz, 1H), 2.66 (dd,  $J = 7.5, 15.0$  Hz, 1H), 3.42 (dddd,  $J = 1.5, 7.0, 10.5, 13.0$  Hz, 1H), 3.75 (dddd,  $J = 2.0, 5.0, 8.0, 11.0$  Hz, 1H), 4.12 (ddd,  $J = 6.0, 10.5, 11.0$  Hz, 2H);  $^{13}\text{C NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  23.7, 27.4, 31.2, 31.5, 31.6, 35.6, 39.0, 50.6, 61.3, 74.7, 178.7, 207.8 [one carbon missing due to overlap]; IR (neat)  $\text{cm}^{-1}$  3420brs, 2929s, 2857s, 2360s, 2341s 1710s; mass spectrum (APCI):  $m/z$  (% relative intensity) 271 ( $\text{M}+\text{H}$ ) $^+$  (100), 253 (71), 213 (85), 169 (50), 151 (89), 133(19), 129 (6), 111 (42);  $m/z$  calcd for  $\text{C}_{15}\text{H}_{26}\text{O}_4\text{Na}^+$  293.1729, found 293.1725.

**Synthesis Homo-Allyl Methyl Ether 13 from (+)-2,3-(*O*)-Iso-Propylidene-*L*-Threitol.**

**TBDPS-Silyl Protection.** To a solution of commercially available (+)-2,3-(*O*)-iso-propylidene-*L*-threitol (1.03 g, 6.30 mmol) in THF (30 mL) was added NaH (252.0 mg, 6.3 mmol) at  $-10\text{ }^{\circ}\text{C}$ . The mixture was gradually warmed up to rt and stirred for 1 h before being cooled back down to  $-10\text{ }^{\circ}\text{C}$  and TBDPSCl (1.77 mL, 6.90 mmol) was added. After 2 h at rt, the reaction was quenched with  $\text{H}_2\text{O}$  (20 mL). The organic solvent was evaporated and the aqueous fraction was extracted with  $\text{CH}_2\text{Cl}_2$  ( $3 \times 20\text{ mL}$ ). The organic phases were combined, washed with sat aq NaCl, dried over  $\text{Na}_2\text{SO}_4$ , filtered, and concentrated under reduced pressure. The crude residue was purified by silica gel flash column chromatography [gradient eluent: 8-25% EtOAc in hexane] to provide the desired TBDPS-silyl ether in 83% yield (2.09 g) as yellow oil.  $R_f = 0.60$  [30% EtOAc in hexane];  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.09 (s, 9H), 1.42 (s, 3H), 1.45 (s, 3H), 2.00 (brds, 1H), 3.68 (dd,  $J = 11.7, 4.4\text{ Hz}$ , 1H), 3.76 (dd,  $J = 10.6, 6.2\text{ Hz}$ , 1H), 3.84 (dt,  $J = 9.7, 4.9\text{ Hz}$ , 2H), 4.03 – 3.95 (m, 1H), 4.12 (ddd,  $J = 12.1, 11.3, 5.7\text{ Hz}$ , 1H), 7.53 – 7.32 (m, 6H), 7.69 (t,  $J = 6.1\text{ Hz}$ , 4H); mass spectrum (ESI):  $m/z$  (% relative intensity) 423.2 ( $\text{M}+\text{Na}$ )<sup>+</sup> (100), 401.1 ( $\text{M}+\text{H}$ )<sup>+</sup> (100);  $m/z$  calcd for  $\text{C}_{23}\text{H}_{32}\text{O}_4\text{SiNa}^+$  423.1968, found 423.1966.

**$\text{SO}_3$ -Pyridine Oxidation.** To a solution of the above silyl ether (2.09 g, 5.20 mmol), anhyd DMSO (7.38 mL, 104.0 mmol) and anhyd  $\text{Et}_3\text{N}$  (3.62 mL, 25.9 mmol) in  $\text{CH}_2\text{Cl}_2$  (21 mL) was added  $\text{SO}_3$ -pyridine (3.31 g, 20.8 mmol) at  $-10\text{ }^{\circ}\text{C}$ . The solution was stirred at  $-10\text{ }^{\circ}\text{C}$  for 2 h and was quenched with  $\text{H}_2\text{O}$  at  $-10\text{ }^{\circ}\text{C}$ . The organic phase was separated and the aqueous fraction was extracted with  $\text{CH}_2\text{Cl}_2$  ( $3 \times 20\text{ mL}$ ). The combined organic phases were dried over  $\text{Na}_2\text{SO}_4$ , filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 10-30% EtOAc in hexane] to provide aldehyde **12** as colorless oil in 82% yield (1.69 g). **12**:  $R_f = 0.35$  [50% EtOAc in hexane];  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.09 (s, 9H), 1.45 (s, 3H), 1.52 (s, 3H), 3.84 (dd,  $J = 4.2, 11.1\text{ Hz}$ , 4H), 3.90 (dd,  $J = 4.5, 11.1\text{ Hz}$ , 1H), 4.22 (dt,  $J = 4.2, 6.9\text{ Hz}$ , 1H), 4.48 (dd,  $J = 1.8, 7.2\text{ Hz}$ , 1H), 7.41-7.48 (m, 6H), 7.69-7.74 (m, 4H), 9.83 (d,  $J = 1.5\text{ Hz}$ , 1H).

**Asymmetric Allylation of Aldehyde 12.** To a solution of (-)-(Ipc)<sub>2</sub>BOMe (1.80 g, 5.69 mmol) in  $\text{Et}_2\text{O}$  (15 mL) was added allylmagnesium bromide (1.0 M in  $\text{Et}_2\text{O}$ , 4.93 mL, 4.93 mmol) at  $0\text{ }^{\circ}\text{C}$ . The solution was warmed up to rt and stirred for an additional 1 h to give a white suspension. The suspension was cooled to  $0\text{ }^{\circ}\text{C}$  and allowed to settle for 0.5 h. The upper supernatant was transferred to a solution of aldehyde **12** (1.51 g, 3.79 mmol) in ether (10 mL) *via* cannula at  $-78\text{ }^{\circ}\text{C}$  and the mixture was stirred at  $-78\text{ }^{\circ}\text{C}$  for 3 h before it was quenched with aq NaOH (3.0 M, 20 mL) and 30%  $\text{H}_2\text{O}_2$  (8 mL) at  $-78\text{ }^{\circ}\text{C}$ . The mixture was reflux overnight. The organic phase was separated and the aqueous fraction was extracted with  $\text{Et}_2\text{O}$  ( $3 \times 20\text{ mL}$ ). The combined organic phases were dried over  $\text{Na}_2\text{SO}_4$ , filtered, and

concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 10-20% EtOAc in hexane] followed by removing isopinocampheol (Ipc-OH) byproduct through Kugelrohr distillation at 50 °C [1.0 mmHg] to provide the pure homoallyl alcohol as colorless oil in 72% yield (1.20 g).  $R_f = 0.60$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = -3.83$  [c 0.31, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 1.09 (s, 9 H), 1.41 (s, 3H), 1.42 (s, 3H), 2.23 (ddd,  $J = 7.5, 7.5, 14.5$  Hz, 1H), 2.41 (m, 1H), 2.52 (d,  $J = 3.0$  Hz, 1H), 3.78 (m, 1H), 3.80 (d,  $J = 4.5$  Hz, 2H), 3.82 (dd,  $J = 7.0, 7.0$  Hz, 1H), 4.08 (ddd,  $J = 4.5, 4.5, 7.0$  Hz, 1H), 5.15 (d,  $J = 10.5$  Hz, 1H), 5.16 (d,  $J = 17.0$  Hz, 1H), 5.92 (dddd,  $J = 7.0, 7.0, 10.5, 17.0$  Hz, 1H), 7.41-7.44 (m, 6H), 7.70-7.72 (m, 4H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 19.2, 26.8, 27.0, 27.1, 37.7, 64.7, 71.4, 79.1, 80.5, 109.0, 118.0, 127.8, 129.9, 132.9, 134.4, 135.7; IR (film) cm<sup>-1</sup> 3470 brs, 3072m, 2933s, 2859m, 1112s; mass spectrum (ESI):  $m/z$  (% relative intensity) 463.2 (M+Na)<sup>+</sup> (100);  $m/z$  calcd for C<sub>26</sub>H<sub>36</sub>O<sub>4</sub>SiNa<sup>+</sup> 463.2281, found 463.2275.

**Methyl Ether Formation.** To a solution of the above homoallylic alcohol (1.20 g, 2.72 mmol) in THF (14 mL) was added NaH (163.2 mg, 4.08 mmol) at -10 °C. The solution was warmed up to rt and stirred for an additional 1 h before MeI (0.34 mL, 5.44 mmol) was added. The mixture was stirred at rt for 12 h and quenched with H<sub>2</sub>O (15 mL). The organic phase was evaporated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 20 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 2-3% EtOAc in hexane] to provide methyl ether **13** as colorless oil in 59% yield (725.0 mg). **13**:  $R_f = 0.70$  [16% EtOAc in hexane];  $[\alpha]_D^{23} = -7.54$  [c 0.93, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 1.10 (s, 9 H), 1.45 (s, 3H), 1.46 (s, 3H), 2.33-2.45 (m, 2H), 3.39 (s, 3H), 3.40 (m, 1H), 3.79 (dd,  $J = 4.0, 11.0$  Hz, 1H), 3.90 (dd,  $J = 3.5, 11.0$  Hz, 1H), 4.08 (dt,  $J = 4.0, 7.0$  Hz, 1H), 4.12 (dd,  $J = 5.0, 8.0$  Hz, 1H), 5.11 (d,  $J = 10.0$  Hz, 1H), 5.15 (dd,  $J = 1.5, 17.0$  Hz, 1H), 5.91 (ddt,  $J = 7.0, 10.0, 17.0$  Hz, 1H), 7.40-7.45 (m, 6H), 7.73-7.76 (m, 4H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 19.3, 26.9, 27.2, 27.3, 34.7, 58.1, 64.8, 77.9, 79.6, 81.5, 109.2, 117.2, 127.7, 129.7, 133.3, 134.6, 135.5; IR (film) cm<sup>-1</sup> 3072m, 2933s, 2861m, 1108s; mass spectrum (ESI):  $m/z$  (% relative intensity) 477.2 (M+Na)<sup>+</sup> (100);  $m/z$  calcd for C<sub>27</sub>H<sub>38</sub>O<sub>4</sub>SiNa<sup>+</sup> 477.2432, found 477.2418.

### Synthesis of Methoxy Aldehyde 14.

**Hydroboration of Methyl Ether 13.** To a solution of methyl ether **13** (91.6 g, 0.20 mmol) in THF (2 mL) was added 9-BBN (0.5 M, 0.81 mL, 0.4 mmol) at 0 °C. The solution was warmed up to rt and stirred for 5 h before aq NaOH (3.0 M, 2 mL) and 30% H<sub>2</sub>O<sub>2</sub> (1 mL) were added. The mixture was refluxed for 2 h. The organic phase was evaporated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 10 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure.

The residue was purified by silica gel flash column chromatography [gradient eluent: 30-40% EtOAc in hexane] to provide the desired alcohol as colorless oil in 71% yield (66.9 mg).  $R_f = 0.30$  [30% EtOAc in hexane];  $[\alpha]_D^{23} = -18.0$  [c 0.75, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  1.07 (s, 9H), 1.42 (s, 3H), 1.43 (s, 3H), 1.60-1.75 (m, 4H), 1.89 (brs, 1H), 3.35 (m, 1H), 3.39 (s, 3H), 3.64 (m, 2H), 3.76 (dd,  $J = 4.5, 10.5$  Hz, 1H), 3.87 (dd,  $J = 3.7, 11.0$  Hz, 1H), 4.01 (ddd,  $J = 4.0, 4.0, 8.0$  Hz, 1H), 4.12 (dd,  $J = 5.0, 7.5$  Hz, 1H), 7.38-7.41 (m, 6H), 7.69-7.72 (m, 4H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  19.3, 26.6, 26.8, 27.1, 27.2, 28.5, 58.1, 62.9, 64.7, 78.0, 79.4, 81.6, 109.1, 127.7, 129.7, 133.3, 135.7; IR (film) cm<sup>-1</sup> 3425brs, 3072m, 2936s, 2864m, 1109s; mass spectrum (ESI):  $m/z$  (% relative intensity) 495.3 (M+Na)<sup>+</sup> (100);  $m/z$  calcd for C<sub>27</sub>H<sub>40</sub>O<sub>5</sub>SiNa<sup>+</sup> 495.2537, found 495.2541.

**SO<sub>3</sub>·Pyridine Oxidation.** To a solution of the above alcohol (66.9 mg, 0.14 mmol), anhyd DMSO (0.20 mL, 2.82 mmol), and anhyd Et<sub>3</sub>N (0.11 mL, 0.79 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (2 mL) was added SO<sub>3</sub>·pyridine (89.8 mg, 0.56 mmol) at -10 °C. The solution was stirred at -10 °C for 2 h and was quenched with H<sub>2</sub>O at -10 °C. The organic phase was separated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 5 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 5-15% EtOAc in hexane] to provide aldehyde **14** as colorless oil in 99% yield (65.5 mg). **14**:  $R_f = 0.70$  [30% EtOAc in hexane];  $[\alpha]_D^{23} = -23.5$  [c 5.56, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  1.08 (s, 9H), 1.42 (s, 6H), 1.92 (dt,  $J = 7.0, 7.5$  Hz, 2H), 2.54 (dt,  $J = 1.5, 7.0$  Hz, 2H), 3.32 (s, 3H), 3.33 (m, 1H), 3.77 (dd,  $J = 4.5, 11.5$  Hz, 1H), 3.87 (dd,  $J = 3.5, 11.0$  Hz, 1H), 4.00 (ddd,  $J = 4.0, 4.0, 7.0$  Hz, 1H), 4.10 (dd,  $J = 5.0, 12.0$  Hz, 1H), 7.39-7.42 (m, 6H), 7.69-7.72 (m, 4H), 9.75 (s, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$  19.3, 22.8, 26.8, 27.1, 27.2, 39.6, 58.0, 64.6, 77.7, 79.5, 80.8, 109.3, 127.7, 129.7, 133.2, 135.7, 202.1; IR (film) cm<sup>-1</sup> 3071m, 2935s, 2861m, 1726s, 1108s; mass spectrum (ESI):  $m/z$  (% relative intensity) 493.2 (M+Na)<sup>+</sup> (100);  $m/z$  calcd for C<sub>27</sub>H<sub>38</sub>O<sub>5</sub>SiNa<sup>+</sup> 493.2386, found 493.2372.

### Synthesis of Diol Enone **16** and Isolation of Bicyclic Acetal **18**.

**Vinyl Grignard Addition to Aldehyde 14.** To a solution of aldehyde **14** (534.3 mg, 1.14 mmol) in Et<sub>2</sub>O (10 mL) was added vinyl magnesium bromide (1.0 M in Et<sub>2</sub>O, 2.28 mL, 2.28 mmol) dropwise at -78 °C. The solution was stirred for 3 h at -78 °C and quenched with sat aq NaHCO<sub>3</sub> (10 mL). The organic phase was separated and the aqueous fraction was extracted with Et<sub>2</sub>O (3 × 10 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 15-20% EtOAc in hexane] to provide allyl alcohol **15** as a mixture of diastereomers in 68% yield (385.2 mg). **15**:  $R_f = 0.30$  [25% EtOAc in hexane]; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.08 (s, 9H), 1.43 (s, 6H), 1.63-1.81 (m, 4H), 1.90 (brs,

1H), 3.33 (m, 1H), 3.38 (s, 3H), 3.75 (dd,  $J = 4.5, 11.1$  Hz, 1H), 3.88 (dd,  $J = 3.3, 11.1$  Hz, 1H), 4.02 (ddd,  $J = 4.2, 4.2, 8.1$  Hz, 1H), 4.12 (dd,  $J = 7.2, 7.2$  Hz, 2H), 5.11 (dd,  $J = 1.2, 10.5$  Hz, 1H), 5.24 (dt,  $J = 17.1, 1.5$  Hz, 1H), 5.88 (dddd,  $J = 1.6, 8.0, 11.5, 17.1$  Hz, 1H), 7.36-7.45 (m, 6H), 7.69-7.74 (m, 4H); mass spectrum (APCI):  $m/z$  (% relative intensity) 499.1 (M+H)<sup>+</sup> (100);  $m/z$  calcd for C<sub>29</sub>H<sub>42</sub>O<sub>5</sub>SiNa<sup>+</sup> 521.2699, found 521.2694.

**Removal of Acetonide.** A solution of the above allylic alcohol (200.0 mg, 0.40 mmol) in mixture of AcOH (5.60 mL) and H<sub>2</sub>O (2.40 mL) was heated to 70 °C for 1.5 h. Then the solution was cooled to rt and sat aq NaHCO<sub>3</sub> was added slowly until pH is about 7. The mixture was diluted with EtOAc (20 mL). Then organic solvents were separated and the aqueous fraction was extracted with EtOAc (3 × 10 mL). The combined organic phases were washed with sat aq NaCl, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 40-60% EtOAc in hexane] to provide the pure triol intermediate in 86% yield (233.8 mg).  $R_f = 0.25$  [50 % EtOAc in hexane]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 1.08 (s, 9H), 1.54-1.80 (m, 4H), 2.75-2.95 (br, 3H), 3.38 (m, 1H), 3.38 (s, 3H), 3.67 (dt,  $J = 6.5, 1.5$  Hz, 1H), 3.77 (dd,  $J = 5.5, 10.0$  Hz, 1H), 3.81 (dd,  $J = 5.5, 10.0$  Hz, 1H), 3.90 (dt,  $J = 1.0, 5.5$  Hz, 1H), 4.13 (m, 1H), 5.09 (d,  $J = 10.5$  Hz, 1H), 5.22 (dd,  $J = 17.5$  Hz, 1H), 5.86 (dddd,  $J = 3.5, 6.0, 9.5, 16.5$  Hz, 1H), 7.38-7.44 (m, 6H), 7.67-7.70 (m, 4H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 19.2, 25.5, 26.9, 32.0, 58.3, 66.2, 69.9, 71.7, 72.8, 82.2, 114.6, 127.9, 129.9, 132.8, 135.6, 141.1; IR (film) cm<sup>-1</sup> 3415brs, 3073m, 2935s, 2861m, 1109s; mass spectrum (APCI):  $m/z$  (% relative intensity) 459.2 (M+H)<sup>+</sup> (100);  $m/z$  calcd for C<sub>26</sub>H<sub>38</sub>O<sub>5</sub>SiNa<sup>+</sup> 481.2386, found 481.2390.

**MnO<sub>2</sub> Oxidation and Formation of Bicyclic Acetal 18.** To a solution of the above triol (0.77 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added MnO<sub>2</sub> (10.0 equiv 7.7 mmol) at rt. The solution was sonicated at rt for 6 h. The mixture was filtered through Celite<sup>TM</sup> and the residue was washed with CH<sub>2</sub>Cl<sub>2</sub> several times. Then the filtrate was collected and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 10-15% EtOAc in hexane] to provide diol enone **16**, which was used right away. To a solution of **16** (7.76 mg, 0.017 mmol) and 1,6-heptadiene-3-ol (3.90 mg, 0.034 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (0.5 mL) was added Tf<sub>2</sub>NH (0.1 M in CH<sub>2</sub>Cl<sub>2</sub>, 0.34 mL, 0.034 mmol) at -78 °C. The solution was stirred at -78 °C for 5 min before quenched with Et<sub>3</sub>N (0.2 mL) at -78 °C. The mixture was warmed to rt and filtered through Celite.<sup>TM</sup> After concentrating the filtrate under reduced pressure, the resulting crude residue (in 81% yield) showed a clean and pure NMR spectrum that could be assigned as bicyclic acetal **18**. **18**:  $R_f = 0.70$  [10% EtOAc in hexane]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 1.07 (s, 9H), 1.93-1.95 (m, 4H), 3.42 (m, 1H), 3.46 (s, 3H), 3.54 (dd,  $J = 9.5, 9.5$  Hz, 1H), 3.66 (dd,  $J =$

5.0, 10.0 Hz, 1H), 3.99 (d,  $J = 5.0, 8.5$  Hz, 1H), 4.65 (m, 1H), 5.20 (d,  $J = 11.0$  Hz, 1H), 5.45 (d,  $J = 17.5$  Hz, 1H), 5.87 (dd,  $J = 10.5, 17.5$  Hz, 1H), 7.38-7.43 (m, 6 H), 7.64-7.66 (m, 4H); mass spectrum (APCI):  $m/z$  (% relative intensity) 439.2 (M+H)<sup>+</sup> (100);  $m/z$  calcd for C<sub>27</sub>H<sub>34</sub>O<sub>4</sub>SiNa<sup>+</sup> 461.2124, found 461.2120.

### Synthesis of Acid 20.

To a solution of aldehyde **14** (4.20 g, 8.90 mmol), 2-methyl-2-butene (4.7 mL, 44.5 mmol), and NaH<sub>2</sub>PO<sub>4</sub> (2.46 g, 19.7 mmol) in the mixture of *t*-BuOH (30 mL) and H<sub>2</sub>O (15 mL) was added NaClO<sub>2</sub> (3.22 g, 35.6 mmol) at -10 °C in 3 portions. The solution was warmed up to rt and stirred for 1 h to give a pale green solution. Then the reaction was quenched with sat aq Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (30 mL). The organic phase was separated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 30 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 30-80% EtOAc in hexane] to provide the carboxylic acid **20** as colorless oil in 98% yield (4.29 g). **20**:  $R_f = 0.40$  [50% EtOAc in hexane];  $[\alpha]_D^{23} = -15.9$  [c 4.37, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 1.08 (s, 9H), 1.42 (s, 6H), 1.86-1.94 (m, 2H), 2.49 (ddd,  $J = 7.0, 16.5, 16.5$  Hz, 1H), 2.52 (ddd,  $J = 7.0, 16.5, 16.5$  Hz, 1H), 3.37 (s, 3H), 3.38 (m, 1H), 3.77 (ddd,  $J = 1.5, 4.5, 11.5$  Hz, 1H), 3.88 (ddd,  $J = 1.5, 4.0, 11.0$  Hz, 1H), 4.02 (m, 1H), 4.11 (m, 1H), 7.38-7.43 (m, 6H), 7.70-7.73 (m, 4H), 11.1 (brs, 1H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 19.3, 21.5, 25.1, 27.1, 27.2, 29.6, 58.2, 64.6, 77.8, 79.5, 80.6, 109.3, 127.7, 129.7, 133.2, 135.7, 179.5; IR (film) cm<sup>-1</sup> 3010brs, 3071m, 2934s, 2861m, 1710s, 1109s; mass spectrum (ESI):  $m/z$  (% relative intensity) 509.2 (M+Na)<sup>+</sup> (100);  $m/z$  calcd for C<sub>27</sub>H<sub>38</sub>O<sub>6</sub>SiNa<sup>+</sup> 509.2330, found 509.2335.

### Synthesis of Pival-Protected Lactone 21.

**Lactone Formation.** To a solution of acid **20** (4.20 g, 8.63 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (45 mL) was added *p*-TsOH-H<sub>2</sub>O (4.92 g, 25.9 mmol) at 0 °C. The solution was warmed up to rt and stirred for 3 h before quenched with sat aq NaHCO<sub>3</sub> (30 mL). The organic phase was separated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 30 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 40-50% EtOAc in hexane] to provide the unprotected lactone as colorless oil in 83% yield (3.08 g).  $R_f = 0.20$  [50% EtOAc in hexane];  $[\alpha]_D^{23} = 48.3$  [c 5.24, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 1.09 (s, 9H), 1.82 (ddd,  $J = 3.9, 6.9, 12.9$  Hz, 1H), 2.04-2.16 (m, 1H), 2.36 (dt,  $J = 18.6, 6.3$  Hz, 1H), 2.55 (ddd,  $J = 17.1, 6.3, 6.3$  Hz, 1H), 3.31 (s, 3H), 3.68 (dd,  $J = 6.0, 10.8$  Hz, 1H), 3.79-3.85 (m, 2H), 3.90-3.97 (m, 1H), 4.46 (d,  $J = 6.0$  Hz, 1H), 7.36-7.37 (m, 6H), 7.70-7.72 (m, 4H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ 19.2, 20.8, 23.2, 26.8, 56.4, 63.9, 70.6, 72.9, 80.0, 127.7, 129.8, 133.0, 135.4, 170.9; IR (film) cm<sup>-1</sup> 3440brs, 3071m, 2936s, 2861m, 1738s, 1110s; mass spectrum (ESI):  $m/z$  (% relative intensity) 451.2

(M+Na)<sup>+</sup> (100); *m/z* calcd for C<sub>24</sub>H<sub>32</sub>O<sub>5</sub>SiNa<sup>+</sup> 451.1911, found 451.1912.

**Pivalation.** To a solution of the above C22-protected lactone (805.7 mg, 1.88 mmol), pyridine (1.6 mL, 18.8 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added (CH<sub>3</sub>)<sub>3</sub>COCl (0.46 mL, 3.8 mmol) followed by DMAP (45.9 mg, 0.38 mmol) at rt. The solution was stirred for overnight and quenched with H<sub>2</sub>O (10 mL). The organic phase was separated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 10 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 10-20% EtOAc in hexane] to provide the pivalate-protected lactone **21** as colorless oil in 77% yield (737.0 mg). **21**: R<sub>f</sub> = 0.65 [50% EtOAc in hexane]; [α]<sub>D</sub><sup>23</sup> = 26.8 [c 3.45, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 1.08 (s, 9H), 1.24 (s, 9H), 2.06 (dddd, *J* = 5.1, 5.7, 9.0, 13.8 Hz, 2H), 2.46 (dt, *J* = 6.0, 17.0 Hz, 1H), 2.69 (ddd, *J* = 6.6, 9.0, 17.1 Hz, 1H), 3.39 (s, 3H), 3.42 (m, 1H), 3.88 (d, *J* = 6.6 Hz, 2H), 4.55 (dd, *J* = 2.4, 6.6 Hz, 1H), 5.33 (dt, *J* = 2.4, 6.6 Hz, 1H), 7.43-7.48 (m, 6H), 7.69-7.74 (m, 4H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ 19.0, 23.1, 26.6, 26.9, 27.0, 38.8, 56.7, 61.4, 71.3, 72.3, 78.6, 127.7, 129.7, 132.7, 135.4, 170.4, 177.5; IR (film) cm<sup>-1</sup> 3071m, 2936s, 2862m, 1741s, 1151s; mass spectrum (APCI): *m/z* (% relative intensity) 513.3 (M+H)<sup>+</sup> (100); *m/z* (ESI) calcd for C<sub>29</sub>H<sub>40</sub>O<sub>6</sub>SiNa<sup>+</sup> 535.2486, found 535.2489.

### Synthesis of Vinyl Ketone and Lactol Mixture **22a/b**.

To a solution of the Piv-protected lactone **21** (131.2 mg, 0.256 mmol) in THF (2 mL) was added vinyl magnesium bromide (1.0 M in THF, 0.51 mL, 0.51 mmol) dropwise at -78 °C. The solution was stirred for 1 h at -78 °C and quenched with sat aq NaHCO<sub>3</sub> (5 mL) and diluted with CH<sub>2</sub>Cl<sub>2</sub> (5 mL). The organic phase was separated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 5 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated under reduced pressure. The residue was purified by silica gel flash column chromatography [gradient eluent: 15-20% EtOAc in hexane] to provide an inseparable mixture of vinyl ketone and lactol **22a/b** as colorless oil in 83% yield (114.7.0 mg) and the recovered starting material (19.5 mg).

**22a/b**: R<sub>f</sub> = 0.65 [50 % EtOAc in hexane]; [α]<sub>D</sub><sup>23</sup> = 9.79 [c 3.36, CHCl<sub>3</sub>]; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 1.05 (s, 9H), 1.25 (s, 9H), 1.45 (d, *J* = 2.7 Hz, 1H), 1.98-2.06 (m, 2H), 2.62-2.80 (m, 2H), 3.17-3.21 (m, 2H), 3.28 (s, 3H), 3.81 (m, 1H), 3.90 (d, *J* = 4.5 Hz, 2H), 5.29 (dt, *J* = 2.5, 7.0 Hz, 1H), 5.83 (d, *J* = 11.5 Hz, 1H), 6.24 (d, *J* = 18.0 Hz, 1H), 6.37 (dd, *J* = 11.0, 18.0 Hz, 1H), 7.40-7.48 (m, 6H), 7.68-7.71 (m, 4H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ 18.9, 22.3, 26.6, 27.1, 33.7, 38.9, 57.3, 64.1, 71.3, 78.9, 103.6, 127.7, 127.7, 128.1, 129.8, 129.8, 132.4, 132.6, 135.4, 135.5, 136.3, 177.6, 201.0; IR (film) cm<sup>-1</sup> 3506brs, 3071m, 2932s, 2858m, 1731s, 1112s; mass spectrum (ESI): *m/z* (% relative intensity) 563.3 (M+Na)<sup>+</sup> (100); *m/z* calcd for C<sub>31</sub>H<sub>44</sub>O<sub>6</sub>SiNa<sup>+</sup> 563.2799, found 563.2810.

**Synthesis of Vinyl Cyclic Acetal 25.**

To a solution of the vinyl ketone and lactol mixture **22a/b** (2.60 mg, 0.0048 mmol) in  $\text{CH}_2\text{Cl}_2$  (0.1 mL) was added MS  $4\text{\AA}$  (10.0 mg), alcohol **24** (4.80 mg, 0.019 mmol) followed by  $\text{Tf}_2\text{NH}$  (0.5 M in toluene, 0.012 mL, 0.0024 mmol) at  $-78\text{ }^\circ\text{C}$ . The solution was stirred at  $-78\text{ }^\circ\text{C}$  for 2 min before quenched with  $\text{Et}_3\text{N}$  (0.05 mL) at  $-78\text{ }^\circ\text{C}$ . The mixture was warmed to rt and filtered through Celite.<sup>TM</sup> After evaporating the solvent under reduced pressure, the resulting crude residue was purified by silica gel flash column chromatography [gradient eluent: 10-25% EtOAc in hexane] to provide the key vinyl cyclic acetal **25** in 56% yield based on starting material recovered. **25**:  $R_f = 0.80$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = 23.5$  [c 0.46,  $\text{CHCl}_3$ ];  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.95 (d,  $J = 6.9$  Hz, 3H), 1.04 (s, 9H), 1.28 (s, 9H), 1.37-1.59 (m, 3H), 1.71-1.81 (m, 2H), 1.91-2.00 (m, 2H), 2.90 (m, 1H), 3.23-3.28 (m, 2H), 3.32 (s, 3H), 3.69-3.75 (m, 2H), 3.81 (m, 1H), 3.83 (s, 3H), 4.04 (m, 1H), 4.07 (d,  $J = 11.5$  Hz, 1H), 4.19 (d,  $J = 11.5$  Hz, 1H), 4.87 (d,  $J = 17.1$  Hz, 1H), 4.92 (d,  $J = 9.9$  Hz, 1H), 5.18 (d,  $J = 12.9$  Hz, 1H), 5.46 (d,  $J = 17.1$  Hz, 1H), 5.59 (m, 1H), 5.67 (m, 1H), 5.78 (dd,  $J = 10.5, 17.1$  Hz, 1H), 6.86 (d,  $J = 8.4$  Hz, 2H), 7.14 (d,  $J = 8.7$  Hz, 2H), 7.38-7.43 (m, 6 H), 7.70-7.72 (m, 4H);  $^{13}\text{C NMR}$  (75 MHz,  $\text{CDCl}_3$ )  $\delta$  15.4, 19.0, 23.0, 26.5, 27.3, 31.5, 34.8, 38.8, 40.3, 55.1, 56.2, 63.7, 66.4, 71.5, 71.8, 73.6, 74.4, 77.1, 97.0, 113.5, 114.4, 115.8, 127.5, 127.6, 129.0, 129.5, 133.3, 135.5, 139.5, 140.5, 158.8, 170.4; IR (film)  $\text{cm}^{-1}$  3071m, 2932s, 2859m, 1731s, 1513m, 1112s; mass spectrum (ESI):  $m/z$  (% relative intensity) 795.6 ( $\text{M}+\text{Na}$ )<sup>+</sup> (100);  $m/z$  calcd for  $\text{C}_{46}\text{H}_{64}\text{O}_6\text{SiNa}^+$  795.4263, found 795.4251.

**Minor Diene 26**:  $R_f = 0.80$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = 27.3$  [c 0.40,  $\text{CHCl}_3$ ];  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  1.07 (s, 9H), 1.24 (s, 9H), 2.16 (ddd,  $J = 3.0, 7.5, 17.7$ , 1H), 2.52 (ddd,  $J = 2.4, 2.4, 17.7$  Hz, 1H), 3.40 (s, 3H), 3.47 (dt,  $J = 5.7, 7.5$  Hz, 1H), 3.90 (d,  $J = 6.0$  Hz, 2H), 4.18 (dd,  $J = 3.6, 7.5$  Hz, 1H), 4.78 (brt,  $J = 4.2$  Hz, 1H), 5.02 (d,  $J = 10.8$  Hz, 1H), 5.38 (d,  $J = 17.1$  Hz, 1H), 5.47 (dt,  $J = 3.6, 6.0$  Hz, 1H), 6.07 (dd,  $J = 10.8, 17.1$  Hz, 1H), 7.30-7.47 (m, 6 H), 7.70-7.73 (m, 4H);  $^{13}\text{C NMR}$  (75 MHz,  $\text{CDCl}_3$ )  $\delta$  19.0, 22.4, 26.4, 26.5, 27.1, 56.5, 61.7, 70.4, 71.6, 74.5, 99.0, 112.7, 127.6, 129.6, 131.3, 133.1, 135.4, 149.9, 170.4; IR (film)  $\text{cm}^{-1}$  3069m, 2960s, 2856m, 1732s, 1279m, 1157s; mass spectrum (ESI):  $m/z$  (% relative intensity) 545.3 ( $\text{M}+\text{Na}$ )<sup>+</sup> (100);  $m/z$  calcd for  $\text{C}_{31}\text{H}_{42}\text{O}_5\text{SiNa}^+$  545.2694, found 545.2691.

**Lactol 27.**

$R_f = 0.32$  [50% EtOAc in hexane];  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  1.08 (s, 9H), 1.85 (m, 1H), 1.91 (m, 1H), 2.60 (ddd,  $J = 6.5, 9.0, 17.0$  Hz, 1H), 2.71 (br, 1H), 2.73 (dddd,  $J = 6.0, 9.0, 17.0, 17.0$  Hz, 1H), 3.28 (s, 3H), 3.38 (ddd,  $J = 4.5, 4.5, 8.5$  Hz, 1H), 3.94 (ddd,  $J = 0.5, 5.0, 11.0$  Hz, 1H), 3.61 (ddd,  $J = 3.5, 3.5, 7.5$  Hz, 1H), 4.01 (ddd,  $J = 1.0, 3.5, 11.5$  Hz, 1H), 4.53 (d,  $J = 11.5$  Hz, 1H), 4.72 (d,  $J = 11.5$  Hz, 1H), 5.80 (dd,  $J = 1.0, 10.5$  Hz, 1H), 6.22 (d,  $J = 17.5$  Hz, 1H), 6.35 (dd,  $J = 10.5, 17.5$  Hz, 1H), 7.27-7.45 (m,

11H), 7.72 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  19.2, 22.7, 26.9, 57.1, 64.2, 70.9, 72.3, 76.9, 79.0, 80.2, 96.5, 127.6, 127.9, 128.4, 129.8, 133.0, 133.2, 135.6, 135.7, 136.6, 138.4; IR (film)  $\text{cm}^{-1}$  3459brs, 3069m, 2932s, 2859s, 1681m, 1428m, 1113s; mass spectrum (APCI):  $m/z$  (% relative intensity) 529.3  $(\text{M}-\text{H}_2\text{O}+\text{H})^+$  (100);  $m/z$  calcd for  $\text{C}_{33}\text{H}_{42}\text{O}_5\text{SiNa}^+$  569.2699, found 569.2697.

### Cyclic Acetal 29.

To a solution of lactol **27** (5.00 mg, 0.0094 mmol) in  $\text{CH}_2\text{Cl}_2$  (0.1 mL) were added MS 4Å (10.0 mg), 3-butene-1-ol (6.77 mg, 0.094 mmol) followed by  $\text{Tf}_2\text{NH}$  (2.64 mg, 0.0094 mmol) at  $-78^\circ\text{C}$ . The solution was stirred at  $-78^\circ\text{C}$  for 5 min before quenched with  $\text{Et}_3\text{N}$  (0.1 mL) at  $-78^\circ\text{C}$ . The mixture was warmed to rt and filtered through Celite<sup>TM</sup>. After evaporation of the solvent under reduced pressure, the resulting crude residue was purified by flash column chromatography on silica gel [Gradient eluent: 2% to 10% EtOAc in hexane] to provide cyclic acetal **29** in 89% yield (5.0 mg).  $R_f = 0.80$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = 28.9$  [c 0.36,  $\text{CHCl}_3$ ];  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  1.06 (s, 9H), 1.45 (ddd,  $J = 4.0$ , 14.0, 14.0 Hz, 1H), 1.74 (ddd,  $J = 4.0$ , 14.0, 24.0 Hz, 1H), 1.88 (ddd,  $J = 4.0$ , 4.0, 14.0 Hz, 1H), 1.95 (dddd,  $J = 4.0$ , 4.0, 8.0, 8.0 Hz, 1H), 2.26 (m, 2H), 3.23 (s, 3H), 3.25 (ddd,  $J = 5.0$ , 10.5, 10.5 Hz, 1H), 3.37 (ddd,  $J = 7.0$ , 7.0, 9.5 Hz, 1H), 3.40 (ddd,  $J = 7.0$ , 7.0, 9.5 Hz, 1H), 3.76 (d,  $J = 9.5$  Hz, 1H), 3.92 (m, 3H), 4.73 (d,  $J = 11.5$  Hz, 1H), 4.79 (d,  $J = 11.5$  Hz, 1H), 5.00 (dd,  $J = 11.0$  Hz, 1H), 5.05 (d,  $J = 17.5$  Hz, 1H), 5.17 (dd,  $J = 1.5$ , 11.0 Hz, 1H), 5.31 (dd,  $J = 2.0$ , 17.5 Hz, 1H), 5.70 (dd,  $J = 11.0$ , 17.5 Hz, 1H), 5.78 (ddt,  $J = 10.5$ , 17.5, 7.0 Hz, 1H), 7.27-7.41 (m, 11H), 7.70 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  18.2, 24.1, 26.9, 34.0, 34.3, 56.0, 60.5, 64.5, 73.1, 73.6, 74.7, 80.9, 97.3, 116.2, 127.2, 127.6, 127.9, 128.2, 129.5, 133.6, 133.8, 135.6, 135.7, 138.8, 139.3; IR (film)  $\text{cm}^{-1}$  3071w, 2929s, 2857s, 1456m, 1104s; mass spectrum (ESI):  $m/z$  (% relative intensity) 623.3  $(\text{M}+\text{Na})^+$  (100);  $m/z$  calcd for  $\text{C}_{37}\text{H}_{48}\text{O}_5\text{SiNa}^+$  623.3169, found 623.3172.

### Cyclic Acetal 30.

$R_f = 0.80$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = 46.1$  [c 0.33,  $\text{CHCl}_3$ ];  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  1.06 (s, 9H), 1.48 (ddd,  $J = 4.0$ , 13.0, 13.0 Hz, 2H), 1.65 (dddd,  $J = 4.0$ , 13.0, 13.0, 13.0 Hz, 2H), 1.81 (m, 2H), 1.95 (m, 2H), 2.29 (ddd,  $J = 6.0$ , 6.0, 6.0 Hz, 1H), 3.15 (ddd,  $J = 4.5$ , 10.5, 10.5 Hz, 1H), 3.21 (s, 3H), 3.39 (m, 5H), 3.49 (dd,  $J = 7.0$ , 14.0 Hz, 1H), 3.68 (d,  $J = 9.5$  Hz, 1H), 3.92 (m, 4H), 4.74 (s, 2H), 5.00 (d,  $J = 10.5$  Hz, 1H), 5.02 (dd,  $J = 1.0$ , 11.0 Hz, 1H), 5.06 (d,  $J = 18.0$  Hz, 1H), 5.07 (dd,  $J = 1.5$ , 17.5 Hz, 1H), 5.79 (ddt,  $J = 10.0$ , 17.0, 7.0 Hz, 2H), 7.27-7.41 (m, 11H), 7.69 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  19.2, 23.9, 26.9, 32.3, 34.2, 34.4, 36.1, 56.0, 59.2, 64.4, 66.7, 70.2, 73.1, 73.7, 74.8, 80.7, 98.0, 116.3, 116.4, 127.2, 127.6, 129.2, 129.5, 133.6, 133.8, 135.3, 135.5, 135.6, 139.2; IR (film)  $\text{cm}^{-1}$  3071m, 2931s, 2858s, 1428m, 1105s; mass spectrum (ESI):  $m/z$  (% relative intensity) 695.6  $(\text{M}+\text{Na})^+$  (100);  $m/z$

calcd for  $C_{41}H_{56}O_6SiNa^+$  695.3744, found 695.3740.

### Diene 31.

$R_f = 0.80$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = -5.87$  [c 0.92,  $CHCl_3$ ];  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$  1.08 (s, 9H), 2.17 (t,  $J = 3.9$  Hz, 2H), 3.40 (s, 3H), 3.74 (ddd,  $J = 3.0, 5.1, 10.2$  Hz, 1H), 3.79 (ddd,  $J = 7.5, 7.5, 7.5$  Hz, 1H), 3.93 (dd,  $J = 5.4, 11.4$  Hz, 1H), 4.01 (dd,  $J = 3.3, 11.4$  Hz, 1H), 4.35 (dd,  $J = 14.5, 6.9$  Hz, 1H), 4.56 (d,  $J = 11.8$  Hz, 1H), 4.71 (dd,  $J = 3.9, 3.9$  Hz, 1H), 4.84 (d,  $J = 11.8$  Hz, 1H), 4.97 (d,  $J = 10.8$  Hz, 1H), 5.40 (d,  $J = 17.1$  Hz, 1H), 6.01 (dd,  $J = 10.8, 17.1$  Hz, 1H), 7.43 (m, 11H), 7.75 (m, 4H);  $^{13}C$  NMR (75 MHz,  $CDCl_3$ )  $\delta$  19.1, 24.0, 26.7, 56.4, 63.4, 71.6, 72.5, 73.9, 78.0, 99.4, 112.5, 127.5, 127.6, 127.8, 128.2, 129.5, 131.6, 133.3, 135.6, 138.3, 148.9; IR (film)  $cm^{-1}$  3070m, 2931s, 2858s, 1428m, 1112s; mass spectrum (APCI):  $m/z$  (% relative intensity) 529.2 ( $M+H$ )<sup>+</sup> (100);  $m/z$  calcd for  $C_{33}H_{40}O_4SiNa^+$  551.2594, found 551.2596.

### BC-Ring 32 via RCM of Cyclic Acetal 25.

To a 0.01 M solution of cyclic acetal **25** (34.8 mg, 0.45 mmol) in toluene was added Grubbs Generation-II Ru-catalyst (0.30 equiv) at rt and the mixture was stirred for 8 h until cyclic acetal **25** was consumed. The suspension was concentrated under reduced pressure and the residue was purified with silica gel flash column chromatography [isocratic eluent: 15% EtOAc in hexane] to provide C11-C23 fragment **32**, colorless oil, in 95% yield. The product yield was based on the amount of cyclic acetal compound **32** and the ratio between cyclic acetal and side product was figured out by  $^1H$  NMR analysis. **32**:  $R_f = 0.50$  [20% EtOAc in hexane];  $[\alpha]_D^{23} = -9.88$  [c 0.24,  $CHCl_3$ ];  $^1H$  NMR (500 MHz,  $CDCl_3$ )  $\delta$  0.84 (d,  $J = 9.0$  Hz, 3H), 1.02 (s, 9H), 1.26 (s, 9H), 1.52-1.75 (m, 4H), 1.85 (m, 1H), 2.00 (m, 2H), 2.94 (ddd,  $J = 2.5, 10.5, 10.5$  Hz, 1H), 3.27 (s, 3H), 3.28-3.34 (m, 2H), 3.51 (ddd,  $J = 5.0, 9.0, 9.0$  Hz, 1H), 3.57 (dd,  $J = 2.5, 9.5$  Hz, 1H), 3.76 (ddd,  $J = 3.5, 10.5, 10.5$  Hz, 1H), 3.79 (s, 3H), 3.92 (d,  $J = 9.0, 11.0$  Hz, 1H), 4.12 (d,  $J = 11.5$  Hz, 1H), 4.16 (d,  $J = 11.5$  Hz, 1H), 5.48 (dd,  $J = 2.5, 10.0$  Hz, 1H), 5.54 (ddd,  $J = 3.0, 3.0, 8.5$  Hz, 1H), 5.61 (dd,  $J = 1.5, 10.0$  Hz, 1H), 6.84 (d,  $J = 8.5$  Hz, 2H), 7.14 (d,  $J = 8.5$  Hz, 2H), 7.32-7.37 (m, 6H), 7.67-7.68 (m, 4H);  $^{13}C$  NMR (125 MHz,  $CDCl_3$ )  $\delta$  16.9, 19.4, 23.9, 26.9, 27.7, 33.3, 33.8, 34.5, 39.2, 55.5, 56.5, 64.3, 67.0, 71.2, 71.9, 72.7, 74.6, 93.5, 113.9, 127.8, 128.6, 129.7, 130.0, 130.9, 134.7, 135.9, 159.3, 177.8; IR (film)  $cm^{-1}$  3070m, 2959s, 2859m, 1731s, 1513m, 1101s; mass spectrum (ESI):  $m/z$  (% relative intensity) 767.4 ( $M+Na$ )<sup>+</sup> (100);  $m/z$  calcd for  $C_{44}H_{60}O_6SiNa^+$  767.3950, found 767.3947.

### Mukaiyama Aldol Reaction

**Silyl Enol Ether Formation Using 11.** To a cooled solution of *i*-Pr<sub>2</sub>NH (41.6  $\mu$ L, 0.294 mmol) in THF (2 mL) was added *n*-BuLi (1.6 M in hexane, 172.0  $\mu$ L, 0.274 mmol) at 0 °C. The solution was stirred for

10 min to give the LDA solution (0.5 M in THF) for next step. To the LDA solution was added pyran **11** *via* cannula at  $-78\text{ }^{\circ}\text{C}$  and the mixture was stirred for 30 min. Chlorotrimethylsilane (35.0  $\mu\text{L}$ , 294 mmol) was then added the lithium enolate and the reaction mixture was stirred for 30 min. The reaction mixture was gradually warmed up to  $0\text{ }^{\circ}\text{C}$  and was quenched with sat aq  $\text{NaHCO}_3$  solution (3 mL). The organic phase was extracted with EtOAc, dried over  $\text{Na}_2\text{SO}_4$ , and concentrated under reduced pressure. The crude trimethylsilyl enol ether was used for the next step without further purification.

**Aldol Addition.** To a cooled solution of the above trimethylsilyl enol ether (0.098 mmol) with either aldehyde **33a** [or **33b**] (0.049 mmol) in  $\text{CH}_2\text{Cl}_2$  (1 mL) was added  $\text{BF}_3\text{-Et}_2\text{O}$  (1.08  $\mu\text{L}$ , 0.062 mmol) at  $-78\text{ }^{\circ}\text{C}$  dropwise. The reaction was stirred at  $-78\text{ }^{\circ}\text{C}$  for 1 h before sat aq  $\text{NaHCO}_3$  (3 mL) was added to quench the reaction. After the mixture was warmed up to room temperature, the organic phase was separated and the aqueous fraction was extracted with  $\text{CH}_2\text{Cl}_2$  ( $3 \times 3$  mL). The combined organic phases were dried over  $\text{Na}_2\text{SO}_4$  and concentrated under reduced pressure. The crude residue was purified with flash silica gel column chromatography [25% EtOAc in hexane] to provide respective Mukaiyama-aldol addition product **34** and **35**. **34:** *For the major isomer:*  $R_f = 0.65$  [30% EtOAc in hexane];  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  0.08 (s, 3H), 0.11 (s, 3H), 0.90 (s, 9H), 0.99 (d,  $J = 7.0$  Hz, 3H), 1.19 (s, 9H), 1.21 (m, 2H), 1.44 (dt,  $J = 3.0, 9.5$  Hz, 2H), 1.49-1.61 (m, 4H), 1.74 (dd,  $J = 2.0, 13.0$  Hz, 2H), 1.84 (m, 1H), 2.37-2.46 (m, 2H), 2.60 (m, 1H), 2.66 (dd,  $J = 8.0, 15.0$  Hz, 1H), 3.20 (brd,  $J = 4.0$  Hz, 1H), 3.38-3.42 (m, 1H), 3.75-3.80 (m, 1H), 3.94 (ddd,  $J = 4.0, 4.0, 8.0$  Hz, 1H), 4.10 (ddd,  $J = 6.0, 11.0, 11.0$  Hz, 1H), 4.13 (ddd,  $J = 7.0, 10.5, 10.5$  Hz, 1H), 4.23 (ddd,  $J = 4.0, 8.0, 12.5$  Hz, 1H), 5.01 (d,  $J = 9.0$  Hz, 1H), 5.02 (d,  $J = 18.0$  Hz, 1H), 5.77 (ddd,  $J = 7.0, 10.0, 17.0$  Hz, 1H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  -4.5, -4.7, 13.2, 17.3, 23.0, 25.5, 26.8, 27.3, 30.7, 30.9, 34.9, 39.6, 39.9, 43.0, 51.0, 60.9, 64.0, 71.8, 74.1, 74.5, 114.9, 140.6, 178.0, 209.6; IR (neat)  $\text{cm}^{-1}$  3751m, 2956s, 2936s, 2870s, 2342s, 1719s, 1702s; mass spectrum (MALDI):  $m/z$  (% relative intensity) 535.5 ( $\text{M}+\text{Na}$ ) $^+$  (100);  $m/z$  calcd for  $\text{C}_{28}\text{H}_{52}\text{O}_6\text{SiNa}^+$  535.3431, found 535.3430. **35:** *For the major isomer:*  $R_f = 0.20$  [25% EtOAc in hexane];  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  1.03 (d,  $J = 7.0$  Hz, 3H), 1.19 (s, 9H), 1.20-1.32 (m, 2H), 1.48-1.61 (m, 6H), 1.73 (q,  $J = 6.5, 13.0$  Hz, 2H), 1.83 (m, 1H), 2.38 (dd,  $J = 5.0, 15.0$  Hz, 1H), 2.56 (m, 2H), 2.64 (dd,  $J = 8.0, 15.0$  Hz, 1H), 3.26 (d,  $J = 3.5$  Hz, 1H), 3.39 (dddd,  $J = 1.0, 6.5, 6.5, 12.5$  Hz, 1H), 3.69 (dddd,  $J = 1.5, 4.0, 4.0, 8.0$  Hz, 1H), 3.75 (m, 1H), 3.81 (s, 3H), 4.11 (ddd,  $J = 7.0, 7.0, 12.5$  Hz, 2H), 4.26 (m, 1H), 4.51 (d,  $J = 10.5$  Hz, 1H), 4.58 (d,  $J = 10.5$  Hz, 1H), 5.03 (ddd,  $J = 1.5, 1.5, 7.5$  Hz, 1H), 5.06 (ddd,  $J = 1.5, 1.5, 17.5$  Hz, 1H), 5.81 (ddd,  $J = 7.5, 10.5, 17.5$  Hz, 1H), 6.87 (d,  $J = 9.0$  Hz, 2H), 7.28 (d,  $J = 8.5$  Hz, 2H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  14.1, 23.5, 27.3, 31.2, 35.4, 37.5, 38.8, 40.4, 50.1, 51.2, 55.4, 61.0, 64.8, 72.2, 74.4, 79.0, 113.9, 114.8, 129.6, 131.0, 140.8, 159.3, 178.6, 210.4; IR (neat)  $\text{cm}^{-1}$  3871m, 3751m, 3677m, 3651m, 3494brs, 2956s, 2936s, 2870s, 2342s, 1719s, 1702s, 1616s; mass spectrum (MALDI):  $m/z$  (% relative intensity)

541(M+Na<sup>+</sup>) (100), 518 (M<sup>+</sup>) (4), 409 (4), 321 (10), 273 (39); *m/z* calcd for C<sub>30</sub>H<sub>46</sub>O<sub>7</sub>Na<sup>+</sup> 541.3141, found 541.3144.

### Directed Reduction for Synthesis of 1,3-Anti-Diols.

To a cooled solution of Me<sub>4</sub>NB(OAc)<sub>3</sub>H (26.0 mg, 0.099 mmol) in anhyd CH<sub>3</sub>CN (0.07 mL) was added anhyd HOAc (0.07 mL) at rt. After stirring for 30 min, the solution was cooled to -30 °C and a solution of the above ketone **34** or **35** (0.017 mmol) in CH<sub>3</sub>CN (0.5 mL) was added via cannula. The mixture was stirred at -30 °C for 6 h before being warmed up to rt and stirred for another 6 h. After which, aq sodium potassium tartrate solution (0.5 M, 1 mL) was added to the reaction followed by sat aq NaHCO<sub>3</sub> (5 mL) and CH<sub>2</sub>Cl<sub>2</sub> (5 mL). The organic phase was separated and the aqueous fraction was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 3 mL). The combined organic phases were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The crude residue was purified by flash silica gel column chromatography to provide the pure respective 1,3-*anti*-diol **36** or **37**. **36** with TBS protection: *R<sub>f</sub>* = 0.40 [25% EtOAc in hexane]; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 0.08 (s, 3H), 0.12 (s, 3H), 0.90 (s, 9H), 0.99 (d, *J* = 6.8 Hz, 3H), 1.20 (s, 9H), 1.41-1.69 (m, 10H), 1.75-1.85 (m, 4H), 2.37 (qdd, *J* = 0.8, 6.8, 12.8 Hz, 1H), 3.46 (dddd, *J* = 1.6, 6.0, 6.0, 13.2 Hz, 1H), 3.59 (dddd, *J* = 2.8, 2.8, 10.0, 10.0 Hz, 1H), 3.88 (br, 1H), 3.93 (ddd, *J* = 4.4, 4.4, 7.2 Hz, 1H), 4.00 (m, 1H), 4.06 (dddd, *J* = 2.8, 2.8, 10.0, 10.0 Hz, 1H), 4.13 (t, *J* = 6.0 Hz, 2H), 5.01 (dd, *J* = 0.8, 10.0 Hz, 1H), 5.02 (dd, *J* = 1.2, 11.6 Hz, 1H), 5.78 (ddd, *J* = 7.2, 10.4, 17.2 Hz, 1H); mass spectrum (MALDI): *m/z* (% relative intensity) 537.5 (M+Na<sup>+</sup>) (23); 515.5 (M+H<sup>+</sup>) (100); *m/z* calcd for C<sub>28</sub>H<sub>54</sub>O<sub>6</sub>SiNa<sup>+</sup> 537.3587, found 537.3690. **37** with PMB protection: *R<sub>f</sub>* = 0.35 [33% EtOAc in hexane]; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 1.03 (d, *J* = 6.5 Hz, 3H), 1.19 (s, 9H), 1.21-1.35 (m, 2H), 1.46-1.62 (m, 6H), 1.65-1.72 (m, 2H), 1.75 (m, 2H), 1.82 (m, 1H), 2.60 (m, 1H), 3.32 (m, 1H), 3.59 (brdt, *J* = 2.0, 7.0 Hz, 1H), 3.68 (dddd, *J* = 3.5, 5.0, 8.5, 12.0 Hz, 1H), 3.80 (s, 3H), 4.02 (m, 1H), 4.08 (m, 1H), 4.12-4.18 (m, 2H), 4.27 (m, 1H), 4.40 (dt, *J* = 7.5, 11.0 Hz, 1H), 4.48 (d, *J* = 11.0 Hz, 1H), 4.57 (d, *J* = 10.5 Hz, 1H), 5.03 (dd, *J* = 1.0, 11.5 Hz, 1H), 5.08 (dd, *J* = 1.5, 17.5 Hz, 1H), 5.41 (q, *J* = 5.0 Hz, 1H), 5.82 (ddd, *J* = 7.0, 10.5, 17.5 Hz, 1H), 6.86 (d, *J* = 8.5 Hz, 2H), 7.28 (d, *J* = 8.5 Hz, 2H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 14.1, 19.9, 25.4, 29.8, 33.7, 34.5, 37.5, 38.5, 39.8, 44.0, 55.0, 62.0, 66.5, 74.0, 75.5, 76.0, 79.9, 114.0, 115.0, 129.8, 131.4, 141.2, 159.9, 179.2; mass spectrum (MALDI): *m/z* (% relative intensity) 543.5 (M+Na<sup>+</sup>) (100); *m/z* calcd for C<sub>30</sub>H<sub>48</sub>O<sub>7</sub>Na<sup>+</sup> 543.3298, found 543.3301.

### Standard Conditions as Described for the Acetonide Formation.

**C13-TBS-Protected C9,11-Anti-Acetonide 38:** *R<sub>f</sub>* = 0.42 [10% EtOAc in hexane]; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 0.09 (s, 6H), 0.89 (s, 9H), 0.99 (d, *J* = 8.5 Hz, 1H), 1.18 (s, 9H), 1.19 (m, 2H), 1.31 (s, 3H), 1.33 (s, 3H), 1.41 (m, 1H), 1.48-1.62 (m, 10H), 1.72-1.83 (m, 3H), 2.33 (m, 1H), 3.74 (m, 1H), 3.46 (m,

1H), 3.76 (ddd,  $J = 5.0, 5.0, 9.5$  Hz, 1H), 3.86 (m, 1H), 4.06 (m, 1H), 3.16 (m, 1H), 4.21 (m, 1H), 5.02 (dd,  $J = 2.5, 22.0$  Hz, 1H), 5.03 (dd,  $J = 1.0, 14.5$  Hz, 1H), 5.79 (ddd,  $J = 9.0, 15.0, 22.0$  Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  0.2, 14.5, 18.4, 23.9, 25.1, 25.6, 26.2, 27.4, 32.1, 35.2, 36.0, 38.9, 39.6, 40.5, 42.9, 43.6, 61.5, 62.9, 64.2, 72.6, 74.1, 74.7, 100.3, 114.8, 140.7, 179.6; mass spectrum (APCI):  $m/z$  (% relative intensity) 555.4 (M+H) $^+$  (100);  $m/z$  calcd for  $\text{C}_{31}\text{H}_{58}\text{O}_6\text{SiNa}^+$  577.3900, found 577.3895.

**Standard TBAF-Desilylation Conditions. 39:**  $R_f = 0.40$  [25% EtOAc in hexane];  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  1.05 (d,  $J = 6.8$  Hz, 3H), 1.19 (s, 9H), 1.34 (s, 3H), 1.35 (s, 3H), 1.50-1.78 (m, 12H), 1.80-1.86 (m, 2H), 2.21 (dd,  $J = 6.8, 6.8$  Hz, 1H), 2.75 (brs, 1H), 3.33 (m, 1H), 3.39 (m, 1H), 3.70 (m, 1H), 4.07-4.12 (m, 3H), 4.18-4.26 (m, 1H), 5.07 (d,  $J = 16.8$  Hz, 1H), 5.08 (d,  $J = 11.2$  Hz, 1H), 5.83 (ddd,  $J = 8.0, 11.6, 15.6$  Hz, 1H); mass spectrum (APCI):  $m/z$  (% relative intensity) 441.1 (M+H) $^+$  (100);  $m/z$  calcd for  $\text{C}_{25}\text{H}_{44}\text{O}_6\text{Na}^+$  463.3036, found 463.3034.

**Standard DDQ Conditions [See Below]. PMP-Acetal 40.**  $R_f = 0.30$  [25% EtOAc in hexane];  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  1.08 (d,  $J = 7.0$  Hz, 3H), 1.19 (s, 9H), 1.40-1.56 (m, 4H), 1.62-1.68 (m, 3H), 1.75-1.79 (m, 2H), 1.80-1.90 (m, 2H), 2.10 (m, 1H), 2.28 (m, 1H), 2.36 (m, 1H), 3.32 (ddd,  $J = 5.5, 5.5, 10.5$  Hz, 1H), 3.45 (m, 1H), 3.61 (m, 1H), 3.79 (s, 3H), 3.86 (m, 1H), 4.04 (ddd,  $J = 5.5, 5.5, 11.0$  Hz, 1H), 4.14 (ddd,  $J = 6.5, 6.5, 11.5$  Hz, 1H), 4.20 (br, 1H), 4.37 (ddd,  $J = 7.0, 7.0, 11.0$  Hz, 1H), 4.53 (q,  $J = 5.0$  Hz, 1H), 5.05 (dd,  $J = 1.0, 9.5$  Hz, 1H), 5.06 (dd,  $J = 1.0, 23.5$  Hz, 1H), 5.72 (s, 1H), 5.91 (ddd,  $J = 8.0, 10.5, 25.0$  Hz, 1H), 6.87 (d,  $J = 9.0$  Hz, 2H), 7.41 (d,  $J = 8.5$  Hz, 2H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  15.5, 23.9, 27.5, 31.6, 36.0, 38.3, 39.0, 42.8, 43.4, 55.6, 61.0, 65.6, 70.1, 74.7, 75.5, 76.0, 94.3, 113.8, 115.1, 127.6, 132.1, 140.6, 160.0, 178.9; mass spectrum (APCI):  $m/z$  (% relative intensity) 519.3 (M+H) $^+$ ;  $m/z$  calcd for  $\text{C}_{30}\text{H}_{46}\text{O}_7\text{Na}^+$  541.3141, found 541.3145.

### Synthesis of Aldehyde 43.

**PMB Deprotected Via DDQ.** To a solution of **32** (130.4 mg, 0.17 mmol) in  $\text{CH}_2\text{Cl}_2 : \text{H}_2\text{O}$  (10:1) was added DDQ (0.25 mmol, 1.5 equiv) at rt. After being stirred for 1.5 h, the mixture was quenched by sat aq  $\text{NaHCO}_3$ . The mixture was extracted with  $\text{CH}_2\text{Cl}_2$ , dried over  $\text{MgSO}_4$ , filtered and concentrated in *vacuo*. The residue was purified by silica gel flash column chromatography [gradient eluent: 33-50% EtOAc in hexane] to provide the desired primary alcohol (94.8 mg, 0.15 mmol) in 88% yield as a colorless oil.  $R_f = 0.30$  [33% EtOAc in hexane];  $[\alpha]_D^{23} = +3.50$  [c 0.84,  $\text{CH}_2\text{Cl}_2$ ];  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  0.87 (d,  $J = 9.0$  Hz, 3H), 1.04 (s, 9H), 1.25 (s, 9H), 1.51-1.69 (m, 3H), 1.72-1.89 (m, 3H), 2.01-2.12 (m, 2H), 2.96 (ddd,  $J = 4.0, 10.0, 10.0$  Hz, 1H), 3.25 (s, 3H), 3.45 (dt,  $J = 2.8, 9.6$  Hz, 1H), 3.61 (t,  $J = 4.4$  Hz, 2H), 3.69 (dd,  $J = 2.8, 9.6$  Hz, 1H), 3.80 (dd,  $J = 4.4, 10.8$  Hz, 1H), 3.88 (dd,  $J = 8.0, 10.8$  Hz, 1H), 5.45 (ddd,

$J = 2.4, 4.4, 7.2$  Hz, 1H), 5.49 (dd,  $J = 2.8, 10.0$  Hz, 1H), 5.64 (dd,  $J = 2.0, 10.0$  Hz, 1H), 7.32-7.37 (m, 6 H), 7.67-7.68 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  16.8, 19.4, 24.0, 27.0, 27.6, 33.7, 34.2, 35.0, 39.2, 56.4, 60.1, 63.8, 71.6, 72.5, 72.7, 74.5, 93.8, 127.8, 128.6, 129.8, 133.9, 134.7, 135.9, 177.9; IR (neat)  $\text{cm}^{-1}$  (neat) 3524w, 3073w, 2958s, 2933s, 2858brs, 2361s, 2342s, 1731s, 1699w, 1160s, 1107s; mass spectrum (MALDI):  $m/z$  (% relative intensity) 647.3 ( $\text{M}+\text{Na}$ ) $^+$  (100);  $m/z$  calcd for  $\text{C}_{36}\text{H}_{52}\text{O}_7\text{SiNa}^+$  647.3375, found 647.3352.

***SO<sub>3</sub>·Pyridine Oxidation.*** To a solution of the above primary alcohol (94.8 mg, 0.15 mmol) in DMSO/ $\text{CH}_2\text{Cl}_2$  were added  $\text{SO}_3\cdot\text{pyridine}$  (96.8 mg, 0.61 mmol) and  $\text{Et}_3\text{N}$  (0.12 mL, 0.76 mmol) sequentially in this order at 0 °C and the resulting mixture was stirred for 2 h at that temperature. The residue was purified by silica gel flash column chromatography [isocratic: 25% EtOAc in hexane] to provide C11-C23 fragment aldehyde **43** (101.2 mg, 0.132 mmol) in 90% yield. **43**:  $R_f = 0.50$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = +7.50$  [c 0.24,  $\text{CH}_2\text{Cl}_2$ ];  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.86 (d,  $J = 7.2$  Hz, 3H), 1.04 (s, 9H), 1.24 (s, 9H), 1.55 (dd,  $J = 4.4, 13.2$  Hz, 1H), 1.61-1.69 (m, 1H), 1.73 (ddd,  $J = 3.2, 3.2, 8.4$ , 1H), 1.98 (m, 1H), 2.07 (m, 1H), 2.37 (ddd,  $J = 3.2, 8.4, 16.0$  Hz, 1H), 2.48 (ddd,  $J = 1.2, 3.6, 16.0$  Hz, 1H), 2.96 (dt,  $J = 4.4, 10.0$  Hz, 1H), 3.24 (s, 3H), 3.67 (dd,  $J = 2.4, 9.6$  Hz, 1H), 3.77 (ddd,  $J = 3.6, 8.4, 9.6$  Hz, 1H), 3.86 (dd,  $J = 4.0, 7.2$  Hz, 2H), 5.47 (ddd,  $J = 2.8, 5.6, 8.0$  Hz, 1H), 5.53 (dd,  $J = 2.8, 10.0$  Hz, 1H), 5.64 (dd,  $J = 1.6, 10.0$ , Hz, 1H), 7.35-7.41 (m, 6H), 7.66-7.69 (m 4H), 9.61 (dd,  $J = 1.2, 3.6$  Hz, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  16.6, 19.5, 23.6, 27.1, 27.7, 33.7, 34.1, 39.2, 46.8, 56.4, 60.1, 69.7, 71.7, 72.0, 74.4, 93.9, 127.8, 128.9, 129.6, 129.8, 133.8, 135.8, 177.7, 201.1; IR (neat)  $\text{cm}^{-1}$  (neat) 3457w, 3074w, 2962s, 2935s, 2860brs, 2724w, 2362w, 2345w, 1733s, 1163s, 1108s; mass spectrum (MALDI):  $m/z$  (% relative intensity) 645.3 ( $\text{M}+\text{Na}$ ) $^+$  (100);  $m/z$  calcd for  $\text{C}_{36}\text{H}_{50}\text{O}_7\text{SiNa}^+$  645.3218, found 645.3250.

#### Assembly of The C1-C23 Fragment **44**.

***Mukaiyama Aldol:*** To a cooled mixture of the silyl enol ether prepared as described above using pyran **11** (0.189 mmol) and aldehyde **43** (101.0 mg, 0.13 mmol) in  $\text{CH}_2\text{Cl}_2$  (1 mL) was added  $\text{BF}_3\cdot\text{Et}_2\text{O}$  (47.3 mL, 0.37 mmol) at -78 °C dropwise. The reaction was stirred at -78 °C for 1 h and then sat aq  $\text{NaHCO}_3$  (3 mL) was added to quench the reaction. After the mixture was warmed up to room temperature, the organic phase was separated and the aqueous fraction was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic phases were dried over  $\text{Na}_2\text{SO}_4$  and concentrated under reduced pressure. The residue was purified with silica gel flash column chromatography [gradient eluent: 2-10% EtOAc in hexane] to provide the desired C1-C23 fragment **44** (62.4 mg, 0.069 mmol) in 62% yield (colorless oil) as a single isomer. **44**:  $R_f = 0.50$  [33% EtOAc in hexane];  $[\alpha]_D^{23} = -15.2$  [c 0.43,  $\text{CH}_2\text{Cl}_2$ ];  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  0.086 (d,  $J = 7.0$  Hz, 3H), 1.02 (s, 9H), 1.18 (s, 9H), 1.24 (s, 9H), 1.40-1.47 (m, 1H), 1.50-1.71 (m, 8H), 1.72-1.80 (m, 4H),

1.86 (m, 1H), 2.02 (m, 1H), 2.10 (dd,  $J = 2.5, 8.0$  Hz, 1H), 2.42 (dd,  $J = 5.5, 16.0$ , 1H), 2.54 (d,  $J = 5.5$  Hz, 1H), 2.65 (dd,  $J = 7.5, 15.5$  Hz, 1H), 2.94 (ddd,  $J = 4.5, 10.0, 10.0$  Hz, 1H), 3.03 (d,  $J = 3.0$  Hz, 1H), 3.23 (s, 3H), 3.41 (dddd,  $J = 1.0, 7.5, 7.5, 7.5$  Hz, 1H), 3.53 (td,  $J = 1.0, 8.0$  Hz, 1H), 3.69 (dd,  $J = 2.0, 9.5$  Hz, 1H), 3.76 (dddd,  $J = 5.5, 5.5, 5.5, 5.5$  Hz, 1H), 3.86 (dd,  $J = 4.0, 11.0$  Hz, 1H), 3.88 (dd,  $J = 4.0, 9.5$  Hz, 1H), 4.12 (brd-t,  $J = 6.5$  Hz, 2H), 4.24 (m, 1H), 5.45 (ddd,  $J = 2.0, 4.0, 4.0$  Hz, 1H), 5.47 (dd,  $J = 2.5, 10.0$  Hz, 1H), 5.63 (d,  $J = 10.0$  Hz, 1H), 7.35-7.41 (m, 6H), 7.66-7.69 (m 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  16.9, 19.5, 23.7, 24.0, 27.1, 27.5, 27.7, 31.5, 31.6, 33.7, 33.9, 35.6, 39.0, 39.2, 39.3, 50.5, 51.2, 56.4, 61.3, 63.9, 64.3, 71.4, 71.7, 72.6, 74.4, 74.5, 74.9, 93.8, 127.8, 128.5, 129.7, 133.9, 134.8, 135.9, 178.0, 178.7, 209.4; IR (neat)  $\text{cm}^{-1}$  2957brs, 2933s, 2860brs, 2361s, 2342s, 1728s, 1157s; mass spectrum (ESI):  $m/z$  (% relative intensity) 915.7 ( $\text{M}+\text{Na}^+$ ) (80), 910.7 (60), 640.6 (50), 563.5 (100);  $m/z$  calcd for  $\text{C}_{51}\text{H}_{76}\text{O}_{11}\text{SiNa}^+$  915.5055, found 915.5050.

### Completion of Assembly of C9,11-Anti-Acetonide 42.

**Directed Reduction.** To a solution of tetramethylammonium triacetoxymethylborohydride (85.1 mg, 0.036 mmol) in anhyd acetonitrile (1.8 mL) and anhyd acetic acid (1.8 mL) at  $-30$  °C was added a solution of  $\beta$ -hydroxy ketone **44** (40.5 mg, 0.045 mmol) in anhyd acetonitrile (1 mL). After the stirring for 12 h, the reaction mixture was quenched with sat aq  $\text{NaHCO}_3$  with an additional stirring of 30 min. The mixture was extracted with  $\text{CH}_2\text{Cl}_2$ . The combined organic layers are dried over  $\text{Na}_2\text{SO}_4$ , filtered, concentrated under reduced pressure and the desired diol **45** [C9,11-anti] was isolated by silica gel flash column chromatography [gradient eluent: 25-50% EtOAc in hexane]. However, the undesired diol **46** [C9,11-syn] was also isolated. The combined yield was 91% with the *anti:syn* ratio being 3:1. The *dr* reflects the isolated ratio. We tried to figure out the ratio according to the crude proton NMR but it was not clear by integration.

C9,11-Anti-Diol **45**:  $R_f = 0.40$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = -7.69$  [c 0.52,  $\text{CH}_2\text{Cl}_2$ ];  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.86 (d,  $J = 9.0$  Hz, 3H), 1.02 (s, 9H), 1.19 (s, 9H), 1.23 (s, 9H), 1.45-1.60 (m, 10H), 1.63-1.70 (m, 3H), 1.72-1.80 (m, 5H), 1.85 (m, 1H), 2.02 (m, 1H), 2.18 (ddd,  $J = 2.0, 5.6, 5.6$  Hz, 1H), 2.96 (ddd,  $J = 4.8, 4.8, 4.8$  Hz, 1H), 3.22 (s, 3H), 3.35 (dddd,  $J = 6.0, 6.0, 11.6, 11.6$  Hz, 1H), 3.56-3.61 (m, 2H), 3.77 (dd,  $J = 2.8, 9.6$  Hz, 1H), 3.87 (d,  $J = 5.2$  Hz, 1H), 3.88 (d,  $J = 6.8$  Hz, 1H), 4.07 (ddd,  $J = 5.6, 5.6, 11.2$  Hz, 1H), 4.14-4.22 (m, 2H), 4.32 (ddd,  $J = 6.0, 6.0, 10.8$  Hz, 1H), 5.40 (ddd,  $J = 2.4, 5.2, 5.2$  Hz, 1H), 5.48 (dd,  $J = 2.4, 10.0$  Hz, 1H), 5.65 (dd,  $J = 2.0, 9.6$  Hz, 1H), 7.34-7.40 (m, 6H), 7.66-7.68 (m, 4H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ )  $\delta$  17.0, 19.5, 21.3, 23.9, 27.1, 27.5, 27.7, 31.5, 31.8, 33.4, 33.8, 35.9, 39.3, 39.5, 43.3, 44.6, 56.3, 60.7, 61.1, 63.6, 65.4, 65.8, 71.4, 71.9, 72.9, 74.6, 74.8, 75.7, 93.9, 127.8, 128.4, 129.8, 134.0, 135.0, 135.9, 178.0, 178.9; IR (neat)  $\text{cm}^{-1}$  2934 brs, 2859brs, 2361s, 2341s, 1728s, 1157s, 1105s; mass spectrum (MALDI):  $m/z$  (% relative intensity) 917.7 ( $\text{M}+\text{Na}^+$ ) (100);  $m/z$  calcd

for  $C_{51}H_{78}O_{11}SiNa^+$  917.5206, found 917.5192.

**C9,11-Syn-Diol 46:**  $R_f = 0.50$  [25% EtOAc in hexane];  $[\alpha]_D^{23} = -4.18$  [c 0.41,  $CH_2Cl_2$ ];  $^1H$  NMR (500 MHz,  $CDCl_3$ )  $\delta$  0.85 (d,  $J = 9.0$  Hz, 3H), 1.02 (s, 9H), 1.19 (s, 9H), 1.24 (s, 9H), 1.40-1.90 (m, 18H), 2.02 (m, 2H), 2.92 (dt,  $J = 6.0, 12.5$  Hz, 1H), 3.24 (s, 3H), 3.46-3.53 (m, 2H), 3.58 (m, 1H), 3.70 (dd,  $J = 3.5, 12.5$  Hz, 1H), 3.91 (brd,  $J = 8.0$  Hz, 2H), 3.97-4.03 (m, 2H), 4.15 (m, 1H), 4.17 (dd,  $J = 15.0, 15.0$  Hz, 2H), 5.47 (br-dd,  $J = 2.4, 10.0$  Hz, 2H), 5.64 (dd,  $J = 2.0, 10.0$  Hz, 1H), 7.33-7.39 (m, 6H), 7.65-7.68 (m, 4H);  $^{13}C$  NMR (125 MHz,  $CDCl_3$ )  $\delta$  17.0, 19.4, 23.5, 23.9, 27.0, 27.4, 27.7, 31.3, 32.0, 34.0, 35.6, 38.9, 39.2, 41.0, 43.0, 56.4, 61.2, 64.1, 68.5, 70.1, 71.4, 71.7, 72.3, 72.6, 74.6, 75.2, 76.9, 79.0, 93.6, 127.7, 128.5, 129.7, 133.9, 134.8, 135.9, 178.5, 178.7; IR (neat)  $cm^{-1}$  2934 brs, 2859brs, 2361s, 2341s, 1728s, 1157s, 1105s; mass spectrum (MALDI):  $m/z$  (% relative intensity) 917.8 ( $M+Na$ ) $^+$  (100);  $m/z$  calcd for  $C_{51}H_{78}O_{11}SiNa^+$  917.5206, found 917.5164.

**Acetonide Protection.** To a solution of C9,11-*anti*-diol **45** (18.3 mg, 0.021 mmol) in acetone (4.8 mL) was added a catalytic amount of pyridinium *p*-toluene sulfonate (1.76 mg, 0.007 mmol) and 2,2-dimethoxypropane (10.0  $\mu$ L, 0.063 mmol) and the resulting mixture was stirred over 1 h. The reaction was quenched by sat aq  $NaHCO_3$  and the organic layer was extracted with  $CH_2Cl_2$ . The combined organic layer was dried over  $Na_2SO_4$ , filtered, and concentrated under reduced pressure. The desired C9,11-*anti*-acetonide **42** was isolated by silica gel flash column chromatography [gradient eluent: 20-50% EtOAc in hexane] in 85% yield (15.7 mg, 0.018 mmol). Also, C9,11-*syn*-diol **46** (7.20 mg, 0.008 mmol) was transformed into the corresponding C9,11-*syn*-acetonide **47** using the same reaction protocol in 90% yield (6.95 mg, 0.007 mmol).

**C9,11-Anti-Acetonide 42.**  $R_f = 0.50$  [20% EtOAc in hexane];  $[\alpha]_D^{23} = -2.70$  [c 0.08,  $CH_2Cl_2$ ];  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  0.84 (d,  $J = 7.2$  Hz, 3H), 1.00 (s, 9H), 1.10 (s, 3H), 1.11 (s, 3H), 1.19 (s, 9H), 1.26 (s, 9H), 1.37-1.60 (m, 8H), 1.60-1.95 (m, 9H), 1.98 (m, 1H), 2.90 (ddd,  $J = 4.4, 9.6, 9.6$  Hz, 1H), 3.17 (s, 3H), 3.32 (ddd,  $J = 2.0, 10.0, 10.0$  Hz, 1H), 3.35-3.39 (m, 1H), 3.44 (m, 1H), 3.57 (dd,  $J = 3.2, 9.6$  Hz, 1H), 3.76 (dd,  $J = 2.8, 11.2$  Hz, 1H), 3.92 (dd,  $J = 9.6, 11.2$  Hz, 1H), 3.95-4.04 (m, 2H), 4.13 (ddd,  $J = 6.8, 6.8, 10.8$  Hz, 1H), 4.22 (ddd,  $J = 6.8, 6.8, 10.4$  Hz, 1H), 5.47 (dd,  $J = 2.4, 10.0$  Hz, 1H), 5.52 (ddd,  $J = 3.2, 3.2, 9.6$  Hz, 1H), 5.58 (dd,  $J = 2.0, 10.0$  Hz, 1H), 7.33-7.39 (m, 6H), 7.64-7.69 (m, 4H);  $^{13}C$  NMR (125 MHz,  $CDCl_3$ )  $\delta$  17.3, 19.4, 23.6, 23.9, 25.3, 26.0, 27.0, 27.5, 27.8, 32.1, 32.3, 33.9, 34.9, 36.0, 39.0, 39.2, 39.8, 42.0, 42.9, 56.2, 61.5, 62.7, 63.7, 64.5, 70.4, 71.9, 72.8, 74.1, 74.6, 74.7, 92.9, 100.1, 127.8, 128.5, 129.8, 133.8, 134.0, 136.0, 177.8, 178.7; IR (neat)  $cm^{-1}$  3454brs, 3048s, 2934s 2858m, 2363s, 2341s, 1661w, 1730s, 1284s, 1159s, 1106s, 1032s, 996s, 935s, 740s, 704s; mass spectrum (MALDI):  $m/z$  (% relative intensity) 957.5 ( $M+Na$ ) $^+$  (100);  $m/z$  calcd for  $C_{54}H_{81}O_{11}SiNa^+$  957.5519, found 957.5507.

**C9,11-Syn-Acetonide 47:**  $R_f = 0.50$  [20% EtOAc in hexane];  $[\alpha]_D^{23} = +2.54$  [c 0.14,  $CH_2Cl_2$ ];  $^1H$  NMR

(500 MHz, CDCl<sub>3</sub>) δ 0.81 (d, *J* = 7.0 Hz, 3H), 0.99 (s, 9H), 1.00 (s, 3H), 1.18 (s, 9H), 1.24 (s, 3H), 1.27 (s, 9H), 1.30-1.57 (m, 9H), 1.60-1.86 (m, 8H), 2.01 (m, 1H), 2.90 (ddd, *J* = 4.5, 10.5, 10.5 Hz, 1H), 3.20 (s, 3H), 3.33-3.42 (m, 2H), 3.37 (dd, *J* = 1.5, 10.0 Hz, 1H), 3.56 (dd, *J* = 3.0, 9.5 Hz, 1H), 3.77 (dd, *J* = 2.5, 10.5 Hz, 1H), 3.94 (m, 1H), 3.98 (dd, *J* = 10.0, 10.0 Hz, 1H), 4.09 (ddd, *J* = 2.0, 9.5, 9.5 Hz, 1H), 4.15 (m, 2H), 5.47 (dd, *J* = 2.0, 10.0 Hz, 1H), 5.58 (br-dd, *J* = 2.0, 10.0 Hz, 2H), 7.32-7.38 (m, 6H), 7.65-7.69 (m, 4H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) δ 17.2, 19.4, 20.2, 23.7, 23.9, 26.9, 27.5, 27.9, 30.4, 31.5, 31.9, 33.8, 35.2, 35.8, 38.2, 39.0, 39.3, 42.0, 43.2, 56.4, 61.5, 64.6, 64.7, 65.7, 69.1, 71.9, 72.5, 74.2, 74.5, 74.6, 92.9, 98.6, 127.8, 128.5, 129.8, 133.8, 134.2, 136.0, 177.0, 177.8; IR (neat) cm<sup>-1</sup> 3454brs, 3048s, 2934s, 2858m, 2363s, 2341s, 1730s, 1284s, 1159s, 1106s, 1032s, 935s, 740s, 703s; mass spectrum (MALDI): *m/z* (% relative intensity) 957.6 (M+Na)<sup>+</sup> (100); *m/z* calcd for C<sub>54</sub>H<sub>81</sub>O<sub>11</sub>SiNa<sup>+</sup> 957.5519, found 957.5516.

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