

2-(3,5-Dimethylphenyl)-5-(4-ethylphenyl)-
1,3,4-oxadiazoleChun-Hui Mao,^{a,b} Qing-Min
Wang,^{a*} Run-Qiu Huang,^a
Li Chen,^a Jian Shang^a and
Hai-Bin Song^a^aState Key Laboratory and Institute of
Elemento-Organic Chemistry, Nankai
University, Tianjin, Weijin Road No 94, Tianjin,
People's Republic of China, and ^bHunan Branch
of National Pesticide R&D South Center,
Changsha 410007, Hunan Province, People's
Republic of China

Correspondence e-mail: chmao@eyou.com

Key indicators

Single-crystal X-ray study

 $T = 293\text{ K}$ Mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$ R factor = 0.059 wR factor = 0.148

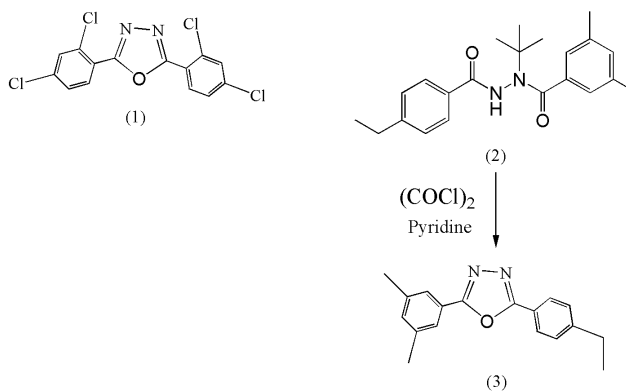
Data-to-parameter ratio = 16.5

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}$, has been synthesized by the reaction of *N*-*tert*-butyl-*N'*-(4-ethylbenzoyl)-3,5-dimethylbenzoylhydrazine with oxalyl chloride. The three rings are nearly coplanar and the molecular geometry is unexceptional.

Comment

Symmetrical 2,5-bis(2,4-dichlorophenyl)-1,3,4-oxadiazole (DCPO), (1), and its analogues have been found to be effective insecticides toward houseflies, faceflies and hornflies (Arrington & Wade, 1980). It has been reported that the oxadiazole ring in DCPO is the biologically active unit (Qian & Zhang, 1996). Recently, synthetic *N*-*tert*-butyl-*N'*-diacylhydrazines (TBDH) have been shown to act as nonsteroidal ecdysone agonists, inducing, especially in Lepidoptera, premature moulting, leading to death (Wing, 1988; Wing *et al.*, 1988; Hsu, 1991; Aller & Ramsay, 1988; Dhadialla *et al.*, 1998). *N*-*tert*-butyl-*N'*-(4-ethylbenzoyl)-3,5-dimethylbenzoylhydrazide (tebufenozide; RH-5992), (2), with its new and selective mode of action, has been the first to be commercialized as an agricultural insecticide to control caterpillar pests by Rohm and Hass (Dhadialla & Jansson, 1999). Research on their quantitative structure-activity relationships has indicated that the substituent groups on the phenyl ring of TBDH play a key role in their larvicidal activities (Oikawa *et al.*, 1994*a,b*; Smagghe *et al.*, 1999; Nakagawa *et al.*, 2001; Nakagawa *et al.*, 1999).



In a search for novel insect-growth regulators, we assembled the active unit of DCPO and the substituent groups on the phenyl ring of RH-5992, to design and synthesize 2-(3,5-dimethylphenyl)-5-(4-ethylphenyl)-1,3,4-oxadiazole, (3), by the reaction of *N*-*tert*-butyl-*N'*-(4-ethylbenzoyl)-3,5-dimethylbenzoylhydrazine, (2), with oxalyl chloride. It has been reported that *N,N'*-diacylhydrazines reacted with SOCl_2 (CIBA Ltd, 1959), or POCl_3 (Shi *et al.*, 2001; Cao *et al.*, 2003) to yield 2,5-disubstituted-1,3,4-oxadiazoles. However, the

Received 1 September 2004

Accepted 6 September 2004

Online 25 September 2004

reactions of *N*-substituted *N,N'*-diacylhydrazines with oxalyl chloride have not hitherto been reported.

The molecular structure of the title compound, (3), is shown in Fig. 1. The title compound contains three ring planes: (I) composed of C11, C12, C13, C14, C15, (II) composed of N1, N2, C10, O1, C1 and (III) composed of C2, C3, C4, C5, C6, C7. The dihedral angles between the (I)/(II), (II)/(III) and (I)/(III) planes are 4.80 (15), 3.46 (16) and 7.49 (15)°, respectively, indicating the near coplanarity of the three rings. A search of the Cambridge Structural Database (Version of November 2003; Allen, 2002) found four comparable 2,5-diphenyl-1,3,4-oxadiazoles: 2-(5-phenyl-1,3,4-oxadiazol-2-yl)-benzoic acid (Smith *et al.*, 1983); 1-(5-phenyl-2-oxazolyl)-2-(5-(2'-methoxy)phenyl-1,3,4-oxadiazol-2-yl)benzene (Doroshenko *et al.*, 2000); bis(*N*-tosyl-L-leucine) 2,5-bis(*o*-aminophenyl)-1,3,4-oxadiazole diamide (Zhao *et al.*, 2000); 2-(4-(4-(*N,N*-bis(2-(acetoxo)ethyl)amino)phenylazo)phenyl)-5-(4-nitrophenyl)-(1,3,4)-oxadiazole (Carella *et al.*, 2002). The bond lengths and angles of the oxadiazole moiety in the title molecule are in good agreement with those in these four structures.

Experimental

Oxalyl chloride (1.27 g, 10.0 mmol) in 1,2-dichloroethane (5 ml) was added to a stirred solution of *N*-*tert*-butyl-*N'*-(4-ethylbenzoyl)-3,5-dimethylbenzoyl hydrazide, 2, (0.70 g, 2.0 mmol) in 1,2-dichloroethane (15 ml) at 273 K. Pyridine (0.79 g, 10.0 mmol) in 1,2-dichloroethane (5 ml) was then added at 273 K. The resulting mixture was stirred at room temperature for 6 h, the excess oxalyl chloride and 1,2-dichloroethane were removed under vacuum, and the residue was diluted with ethyl acetate (20 ml); the organic layer was washed successively with a solution of sodium bicarbonate, water, and brine, and dried over anhydrous sodium sulfate. After evaporation of the solvent, the residue was recrystallized from water and ethanol (2:3) to yield the title compound.

Crystal data

| | |
|----------------------------|--------------------------------------|
| $C_{18}H_{18}N_2O$ | Mo $K\alpha$ radiation |
| $M_r = 278.34$ | Cell parameters from 911 reflections |
| Orthorhombic, <i>Pbca</i> | $\theta = 2.9\text{--}21.8^\circ$ |
| $a = 8.534$ (4) Å | $\mu = 0.08$ mm $^{-1}$ |
| $b = 16.122$ (6) Å | $T = 293$ (2) K |
| $c = 22.481$ (9) Å | Prism, colorless |
| $V = 3093$ (2) Å 3 | $0.38 \times 0.30 \times 0.20$ mm |
| $Z = 8$ | |
| $D_x = 1.195$ Mg m $^{-3}$ | |

Data collection

| | |
|--|--|
| Bruker SMART 1000 CCD area detector diffractometer | 3142 independent reflections |
| φ and ω scans | 1715 reflections with $I > 2\sigma(I)$ |
| Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 1996) | $R_{\text{int}} = 0.046$ |
| $T_{\text{min}} = 0.825$, $T_{\text{max}} = 0.990$ | $\theta_{\text{max}} = 26.4^\circ$ |
| 12360 measured reflections | $h = -10 \rightarrow 4$ |
| | $k = -19 \rightarrow 16$ |
| | $l = -25 \rightarrow 28$ |

Refinement

| | |
|---------------------------------|---|
| Refinement on F^2 | $w = 1/[\sigma^2(F_o^2) + (0.0592P)^2 + 0.3169P]$ |
| $R[F^2 > 2\sigma(F^2)] = 0.059$ | where $P = (F_o^2 + 2F_c^2)/3$ |
| $wR(F^2) = 0.148$ | $(\Delta/\sigma)_{\text{max}} = 0.001$ |
| $S = 1.06$ | $\Delta\rho_{\text{max}} = 0.17$ e Å $^{-3}$ |
| 3142 reflections | $\Delta\rho_{\text{min}} = -0.13$ e Å $^{-3}$ |
| 191 parameters | |
| H-atom parameters constrained | |

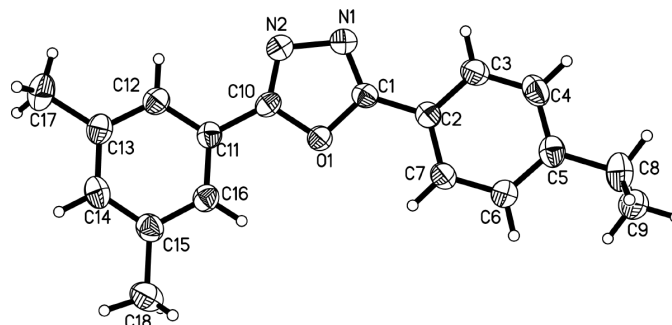


Figure 1

The structure of (3), showing 40% probability displacement ellipsoids and the atom-numbering scheme.

Table 1

Selected geometric parameters (Å, °).

| | | | |
|-------------|-------------|----------------|-----------|
| N1—C1 | 1.291 (3) | O1—C1 | 1.370 (3) |
| N1—N2 | 1.409 (3) | C1—C2 | 1.449 (3) |
| N2—C10 | 1.286 (3) | C10—C11 | 1.456 (3) |
| O1—C10 | 1.365 (3) | | |
| C1—N1—N2 | 106.48 (19) | O1—C1—C2 | 118.9 (2) |
| C10—N2—N1 | 106.60 (19) | N2—C10—O1 | 112.1 (2) |
| C10—O1—C1 | 103.15 (18) | N2—C10—C11 | 129.0 (2) |
| N1—C1—O1 | 111.7 (2) | O1—C10—C11 | 118.9 (2) |
| N1—C1—C2 | 129.4 (2) | | |
| O1—C1—C2—C7 | 2.6 (3) | O1—C10—C11—C16 | −4.7 (3) |

H atoms were placed in calculated positions, with C—H = 0.93, 0.96 or 0.97 Å, and included in the final cycles of refinement using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{parent atom})$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

This work was supported by the National Key Project for Basic Research (2003CB114400), the National Natural Science Foundation of China (20202005), the Research Fund for the Doctoral Program of Higher Education (20010055006) and the Foundation for the Author of National Excellent Doctoral Dissertation of the People's Republic of China (200255).

References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
- Aller, H. E. & Ramsay, J. R. (1988). *Brighton Crop Prot. Conf. Pests Dis.* **2**, 511–518.
- Arrington, J. P. & Wade, L. L. (1980). US Patent, No. 4215129.
- Bruker (1998). *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1999). *SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cao, S., Qian, X. H., Song, G. H., Chai, B. & Jiang, Z. S. (2003). *J. Agric. Food Chem.* **51**, 152–155.
- Carella, A., Castaldo, A., Centore, R., Fort, A., Sirigu, A. & Tuzi, A. (2002). *J. Chem. Soc. Perkin Trans. 2*, pp. 1791–1795.
- CIBA Ltd. (1959). GB Patent No 892767.
- Dhadialla, T. S., Carlson, G. R. & Le, D. P. (1998). *Annu. Rev. Entomol.* **43**, 545–569.
- Dhadialla, T. S. & Jansson, R. K. (1999). *Pest. Sci.* **55**, 357–359.

- Doroshenko, A. O., Kyrychenko, A. V., Baumer, V. N., Verezubova, A. A. & Ptyagina, L. M. (2000). *J. Mol. Struct.* **524**, 289–296.
- Hsu, A. C.-T. (1991). *Synthesis and chemistry of agrochemicals II*, edited by B. R. Baker, J. G. Fenyes and W. K. Moberg. ACS Symposium Series 443, American Chemical Society, Washington, DC, pp. 478–490.
- Nakagawa, Y., Smagghe, G., Kugimiya, S., Hattori, K., Ueno, T., Tirry, L. & Fujita, T. (1999). *Pest. Sci.* **55**, 909–918.
- Nakagawa, Y., Smagghe, G., Paeme, M. V. & Tirry, L. (2001). *Pest Manag. Sci.* **57**, 858–865.
- Oikawa, N., Nakagawa, Y., Nishimura, K., Ueno, T. & Fujita, T. (1994a). *Pest. Sci.* **41**, 139–148.
- Oikawa, N., Nakagawa, Y., Nishimura, K., Ueno, T. & Fujita, T. (1994b). *Pest. Biochem. Physiol.* **48**, 135–144.
- Qian, X. H. & Zhang, R. (1996). *J. Chem. Tech. Biotechnol.* **67**, 124–130.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Shi, W., Qian, X. H., Zhang, R. & Song, G. H. (2001). *J. Agric. Food Chem.* **49**, 124–130.
- Smagghe, G., Nakagawa, Y., Carton, B., Mourad, A. K., Fujita, T. & Tirry, L. (1999). *Arch. Insect Biochem. Physiol.* **41**, 42–53.
- Smith, G., Kennard, C. H. L. & Katekar, G. F. (1983). *Aust. J. Chem.* **36**, 2455–2463.
- Wing, K. D. (1988). *Science*, **241**, 467–469.
- Wing, K. D., Slawecki, R. A. & Carlson, G. R. (1988). *Science*, **241**, 470–472.
- Zhao, H., Wei, Y. & Hua, W. (2000). *J. Mol. Struct.* **553**, 109–115.